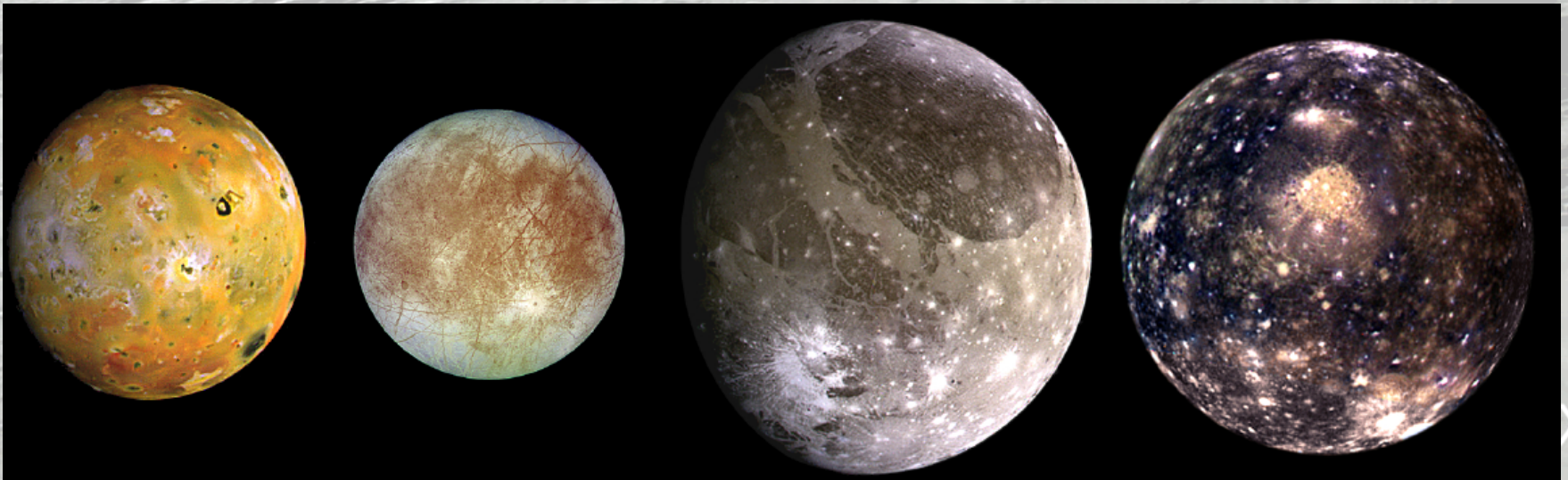


実験惑星学3: ノンウォーターアイスの物性研究

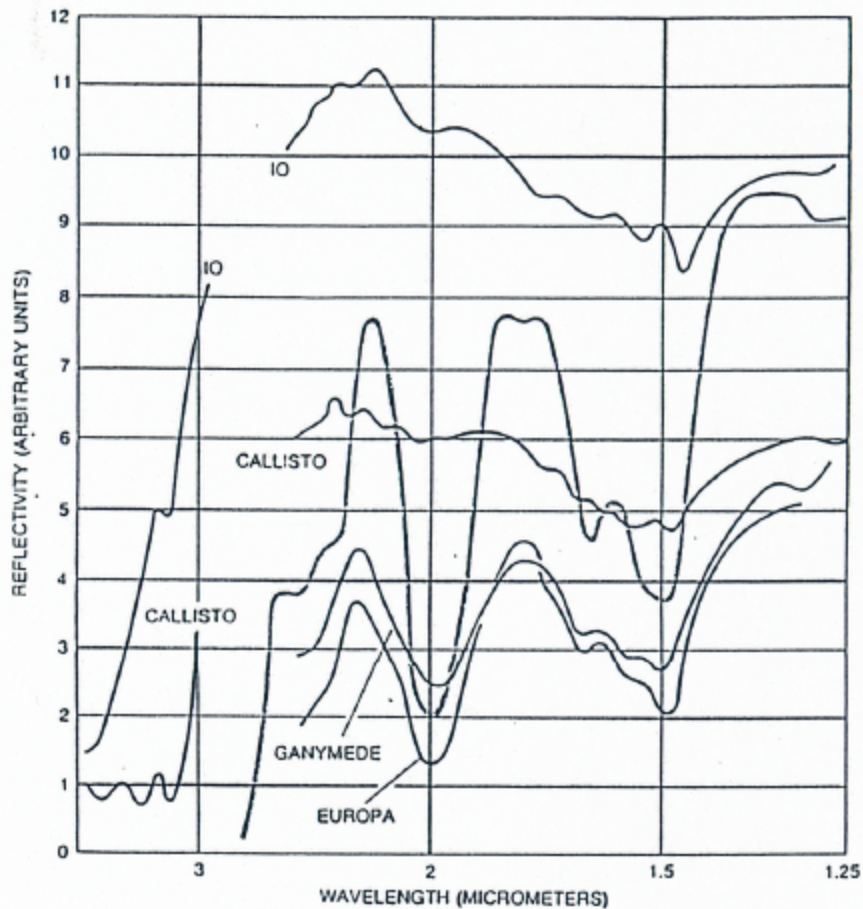
- 外惑星系に普遍的に存在する窒素、メタン、二酸化炭素の氷
- 作成法も未知のものが多い
- 音速や相図が何とか作られているが、確かめるべき
- クラスレート化合物も視野に入れる



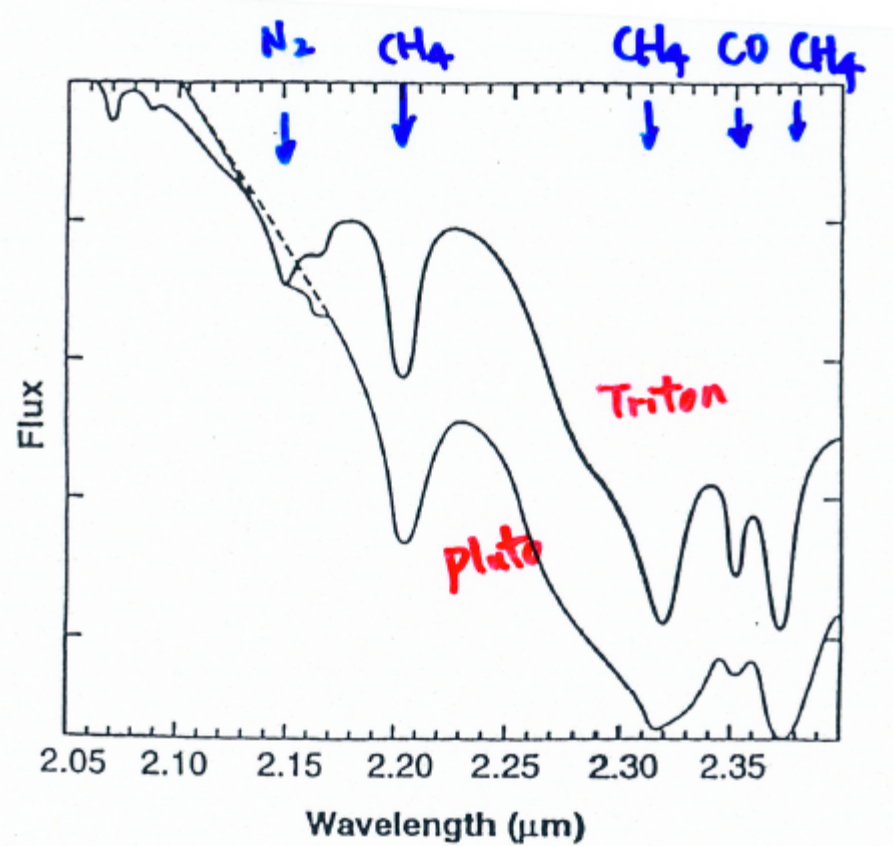
Ices in Solar System

Object	Radius(km)	Average density (g/cm ³)	Albedo	Varieries of ices identified or inferrerd
Io	1815	3.57	0.6	SO ₂
Europa	1569	2.97	0.6	H ₂ O(I)
Ganymede	2631	1.94	0.4	H ₂ O(I,II,V,VI,VII,amorphous in polar caps ?)
Callisto	2400	1.86	0.2	H ₂ O(I,II,V,VI,VII)
Rings of Saturn			0.2 - 0.6	H ₂ O(I,amorphous?)
Mimas	197	1.17	0.77	H ₂ O(I)
Enceladus	251	1.24	1.04	H ₂ O(I,formerly amorphous?) clathrate hydrate ? ammonia hydrate ?
Tethys	524	1.26	0.80	H ₂ O(I)
Dione	559	1.44	0.55	H ₂ O(I)
Rhea	764	1.33	0.65	H ₂ O(I,II)
Titan	2575	1.881	0.2	H ₂ O(I,II,V,VI,VII) clathrate hydrate ?
Hyperion	175*120*100		0.25	H ₂ O(I)
Iapetus	718	1.21	0.5 / 0.04	H ₂ O(I)
s10 - s17	10 - 100		0.5 - 0.9	H ₂ O(I) ?
Rings of Uranus			- 0.03	H ₂ O(I) ? modified CH ₄ ?
Miranda	235	1.35	0.22	H ₂ O(I) CH ₄ ?
Ariel	580	1.66	0.38	H ₂ O(I)
Umbriel	585	1.51	0.16	H ₂ O(I)
Titania	790	1.68	0.23	H ₂ O(I)
Oberon	760	1.58	0.20	H ₂ O(I)
Triton	1350	2.075	0.6 - 0.9	CH ₄ N ₂ CO CO ₂ H ₂ O(I) ? clathrate hydrate ?
Pluto	1145	1.84	0.612	CH ₄ N ₂ CO H ₂ O(I)? clathrate hydrate ?
Charon	642		0.424	H ₂ O(I)

Non-Water Ices
in Triton, Pluto

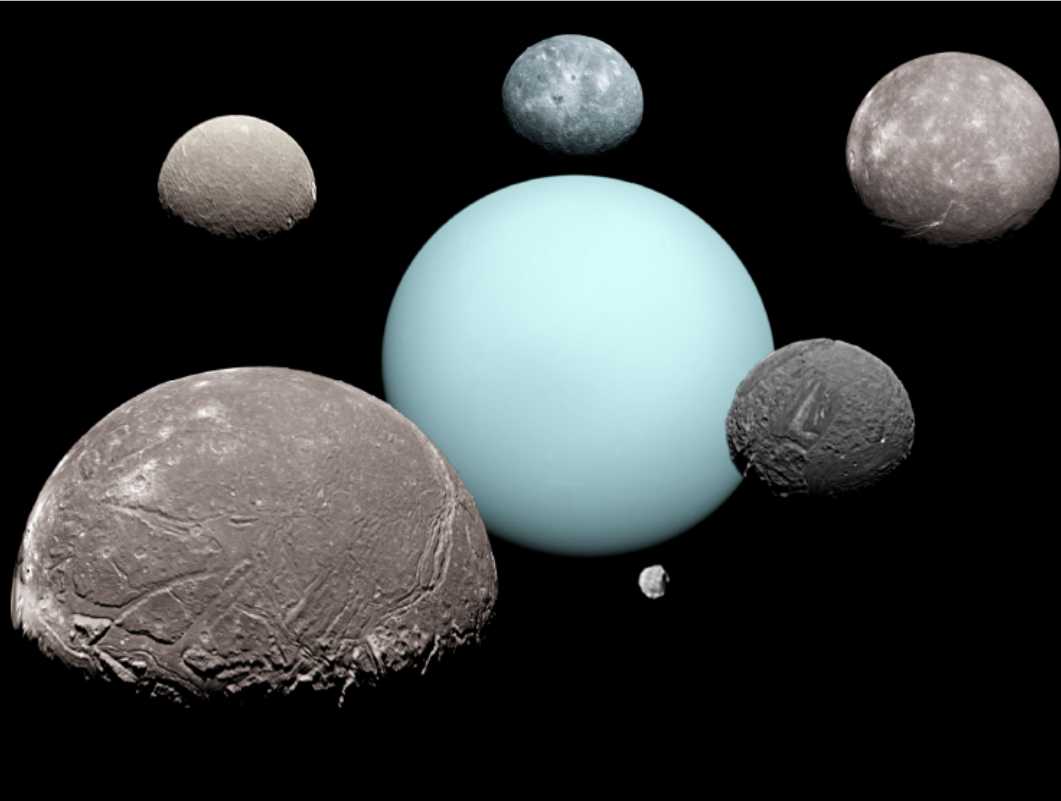
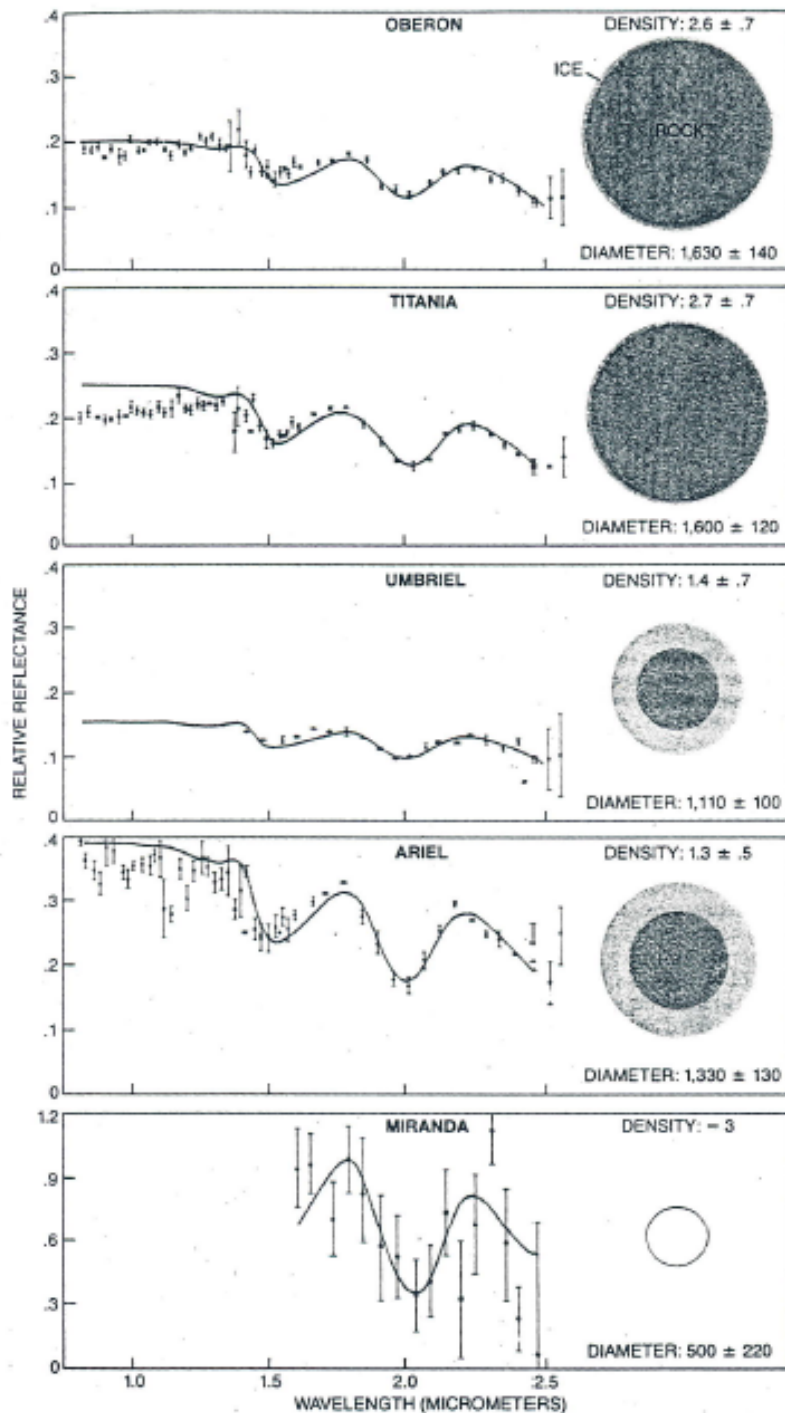


Galileo Satellites



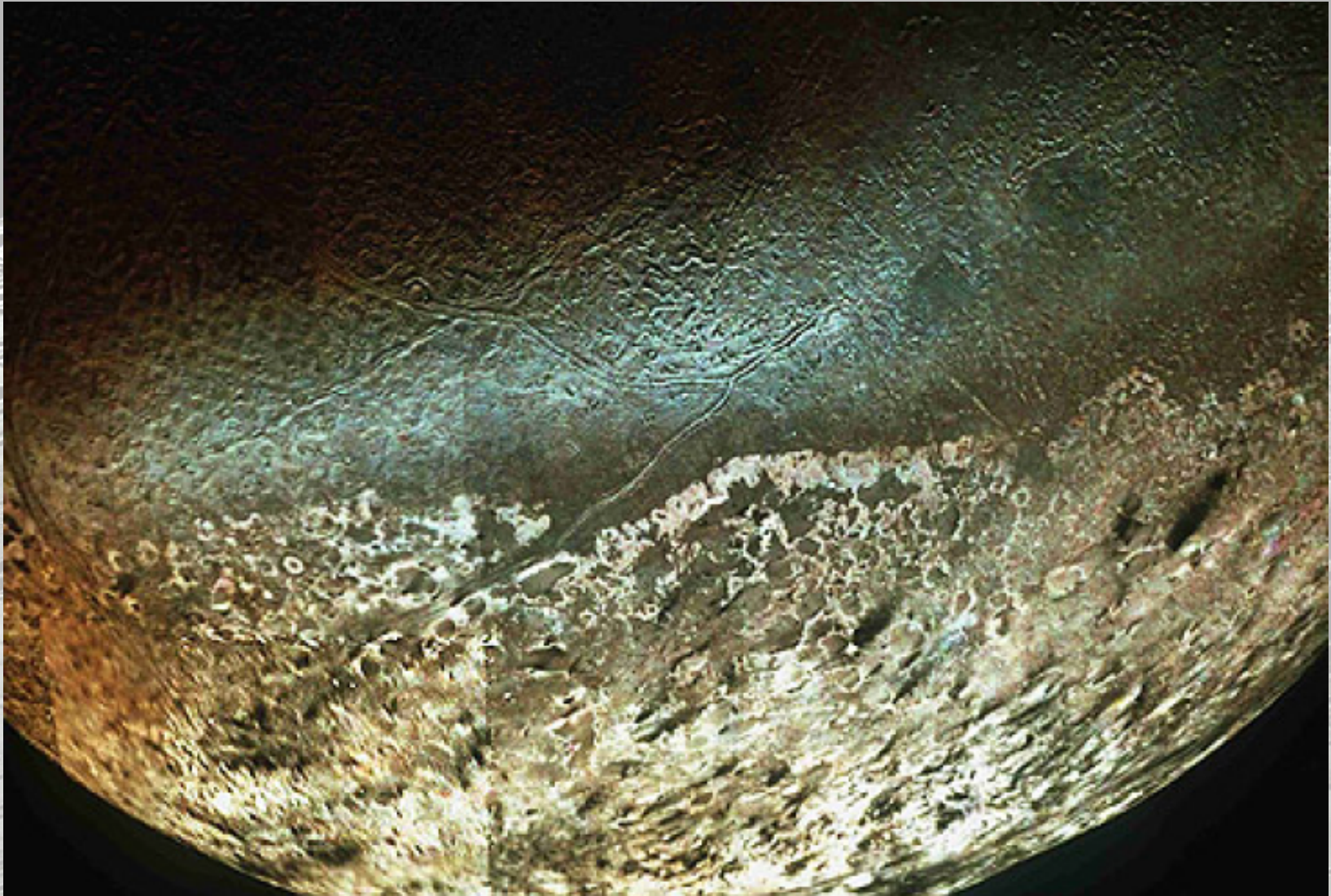
	N ₂	CH ₄	CO	CO ₂	H ₂ O
Triton	90%	0.1%	<0.1%	10%	-----
Pluto	>95%	2%	0.5%	-----	-----
Charon	-----	-----	-----	-----	>95%

Reflective Spectrum of Uranian Satellites



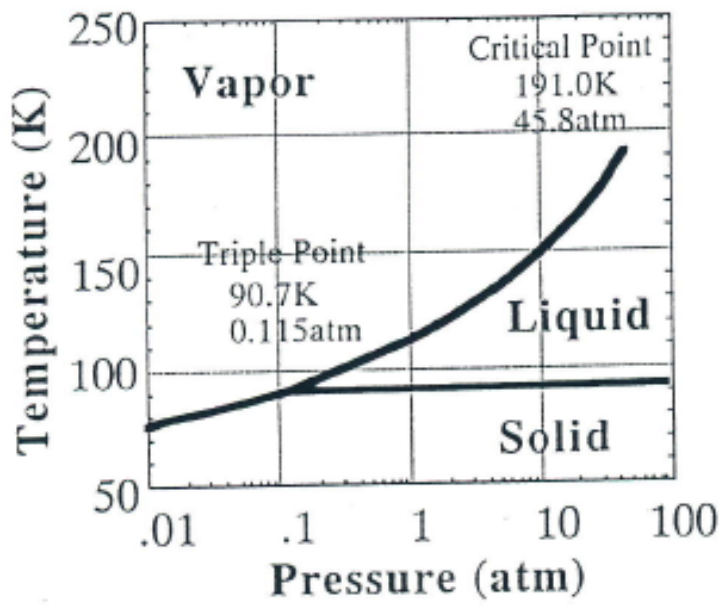
COMPOSITION OF URANIAN MOONS can be inferred from their near-infrared reflectance spectra and from their mean densities. The spectra of the four outer moons all have deep absorption bands at wavelengths of 1.5 and 2.0 micrometers, both of which are characteristic of water ice. The data for Miranda are crude (the vertical lines show the error range) but a strong absorption band is evident at 2.0 micrometers. The satellite spectra (dots) are matched closely by the spectra of two-component laboratory models (lines) consisting of water frost and charcoal. All the moons are thought to have an icy surface and a rocky core, but the differences in mean density (measured in grams per cubic centimeter) suggest that Ariel and Umbriel have thicker ice layers than Titania and Oberon. Miranda's density is too uncertain to allow a conclusion about its internal structure. The diameters are given in kilometers. In comparison, the earth's moon has a diameter of 3,480 kilometers.

Triton by Voyager 2



Cantaloup Terrain

Phase Diagrams of various Non-Water Ices



Phase diagram of CH₄

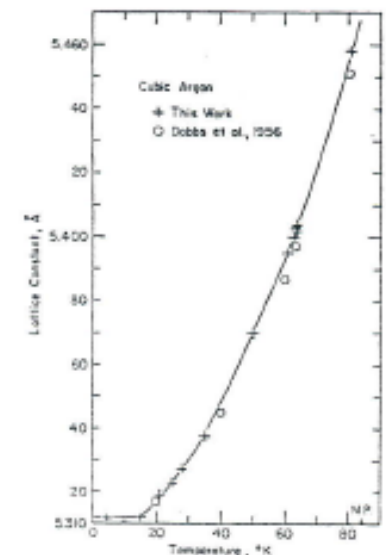
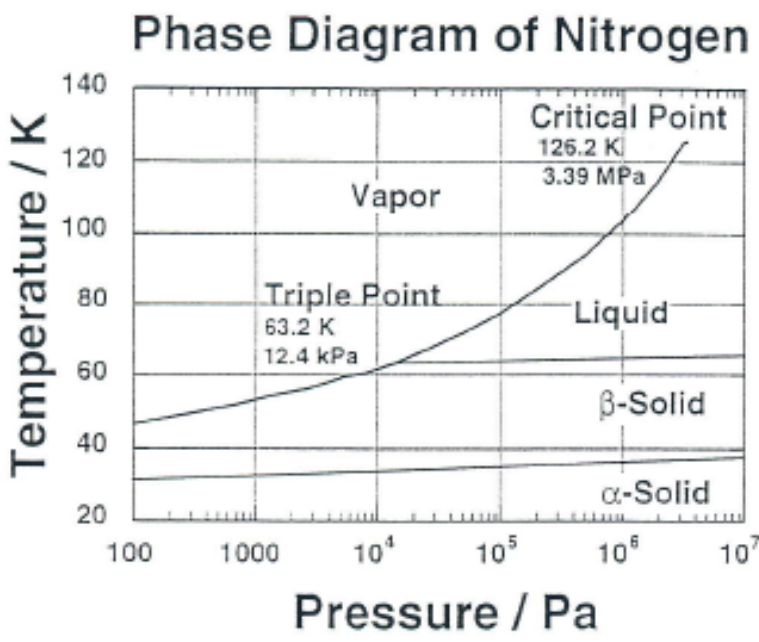


FIG. 1. Lattice constants for face-centered-cubic argon.

Ar

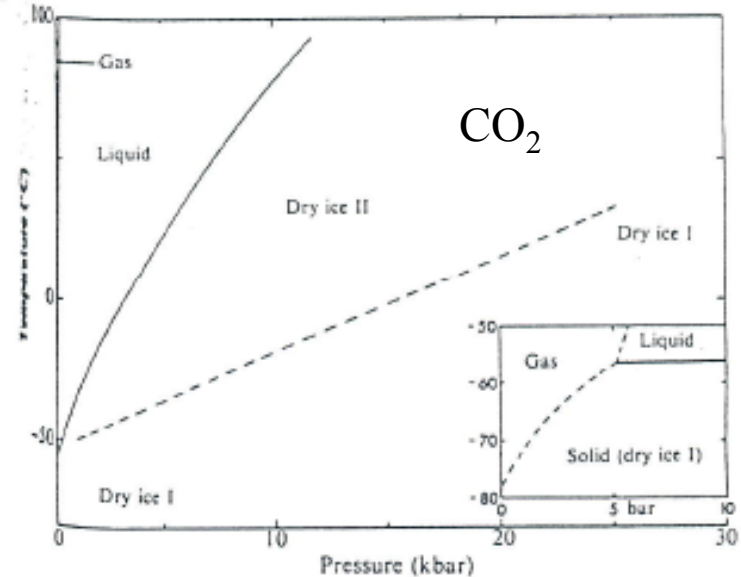
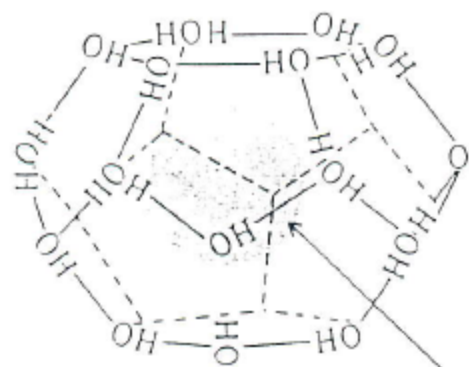
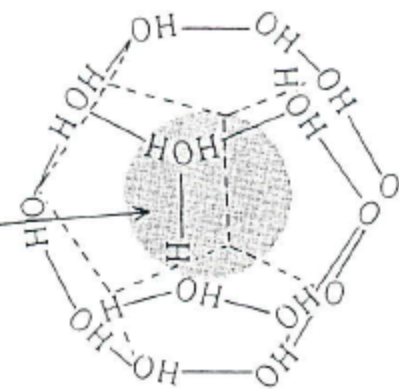


Fig. 1 A tentative P - T phase diagram for CO₂ based on the results of this study and those reported in the literature^{1,2,7,13}. The suggested small dT/dP slope for the phase boundary between dry ice II and dry ice I is consistent with the small density difference between the two phases observed in the present study.

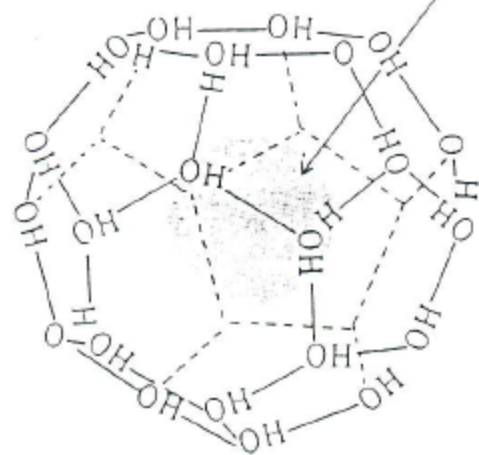
14面体



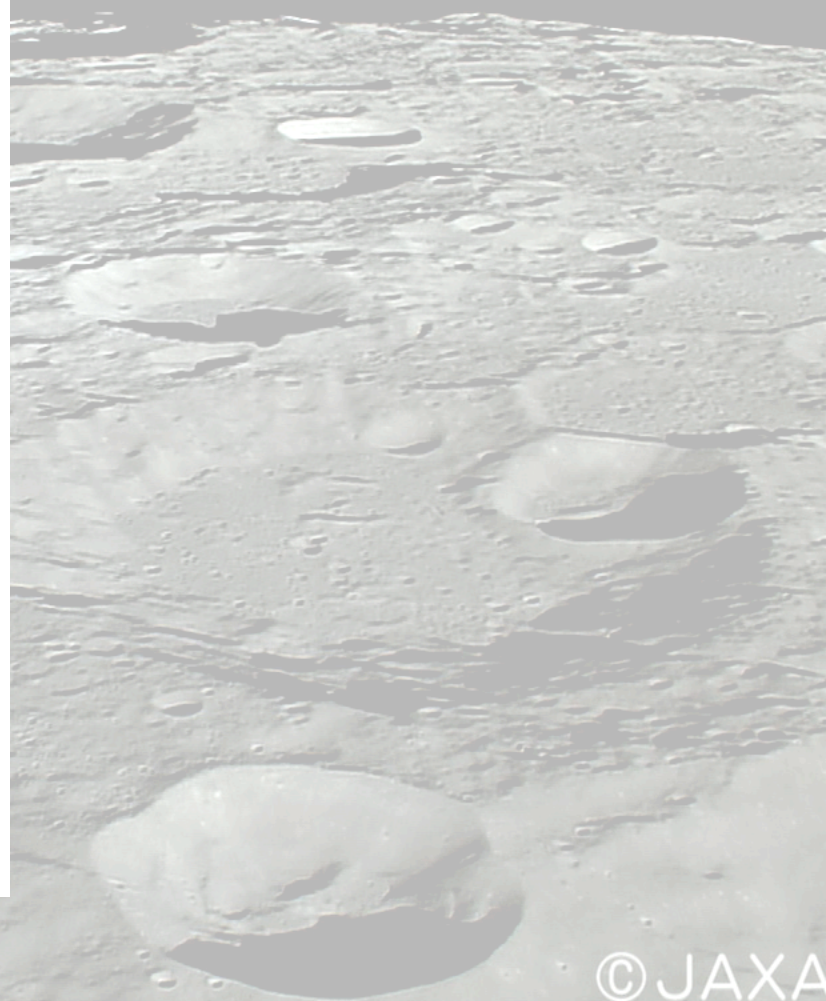
気体分子



12面体



16面体



Phase Diagrams of $\text{CH}_4\text{-N}_2$, $\text{NH}_3\text{-H}_2\text{O}$

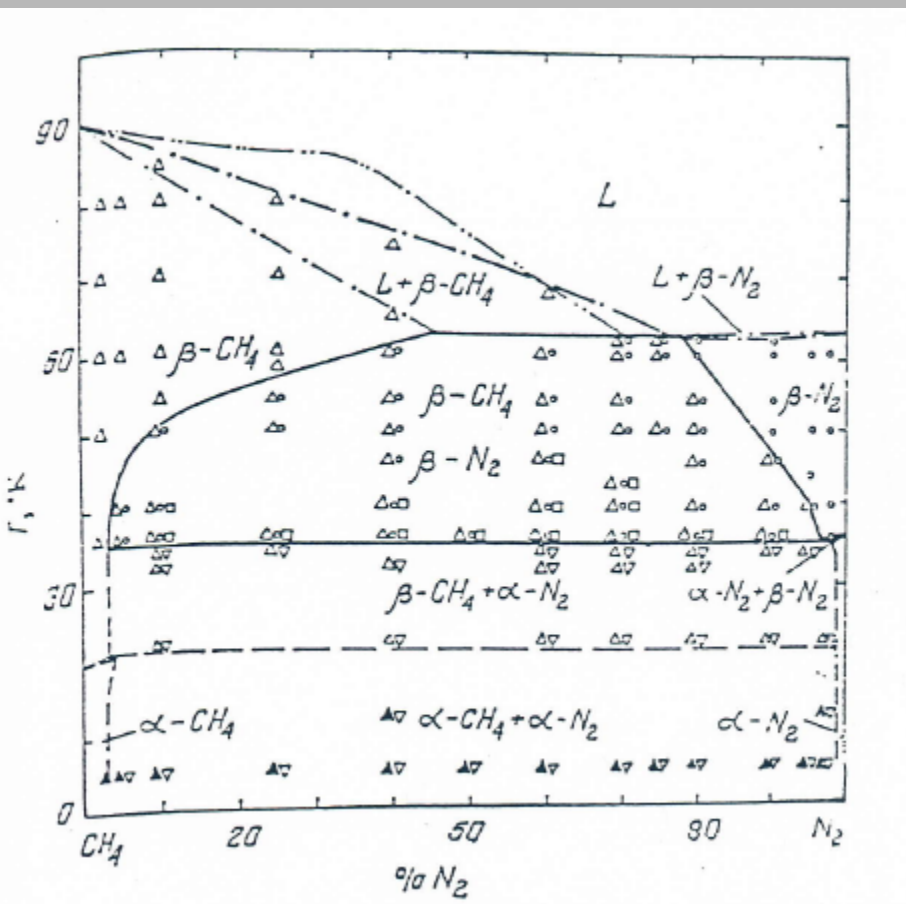


FIG. 1. Phase diagram of solid methane-nitrogen mixtures. The dashed-double dotted line corresponds to the experimental data of Ref. 13 on the melting diagram; the dashed-double dotted line shows the data of Ref. 12 the solid lines are the phase boundaries of the solid mixture phase diagram determined in the present paper. The following are the structures of the phases observed: $\blacktriangle$, ∇) cubic phases of $\alpha\text{-CH}_4$ and $\alpha\text{-N}_2$; Δ , $\square$) fcc lattice of $\beta\text{-CH}_4$ and of the orientationally disordered phase of N_2 ; $\circ$) hcp lattice of $\beta\text{-N}_2$.

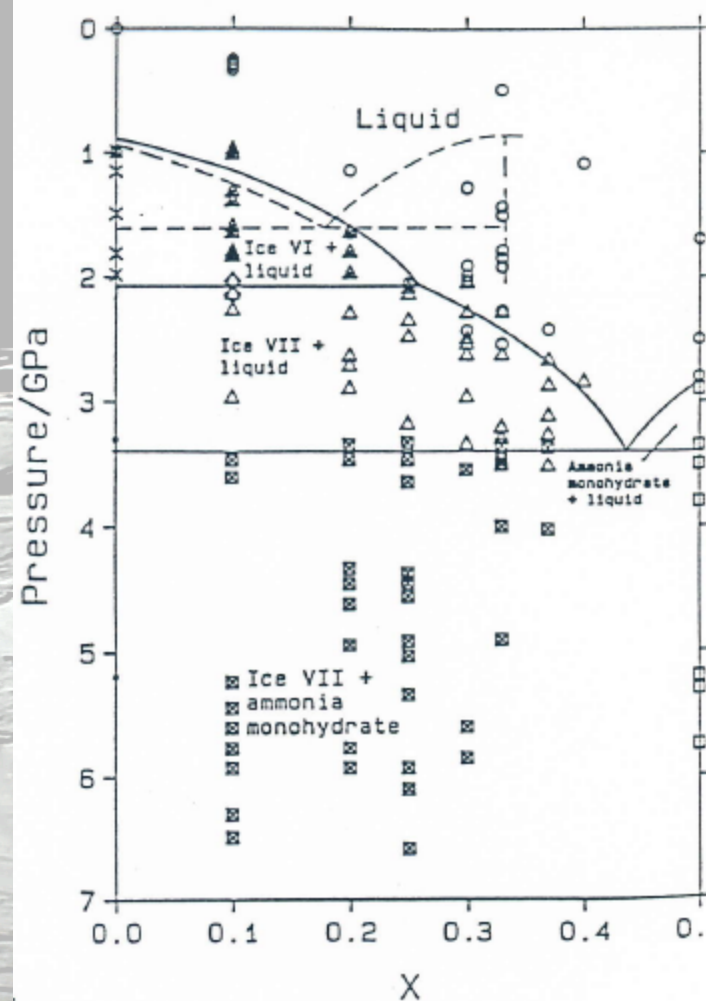
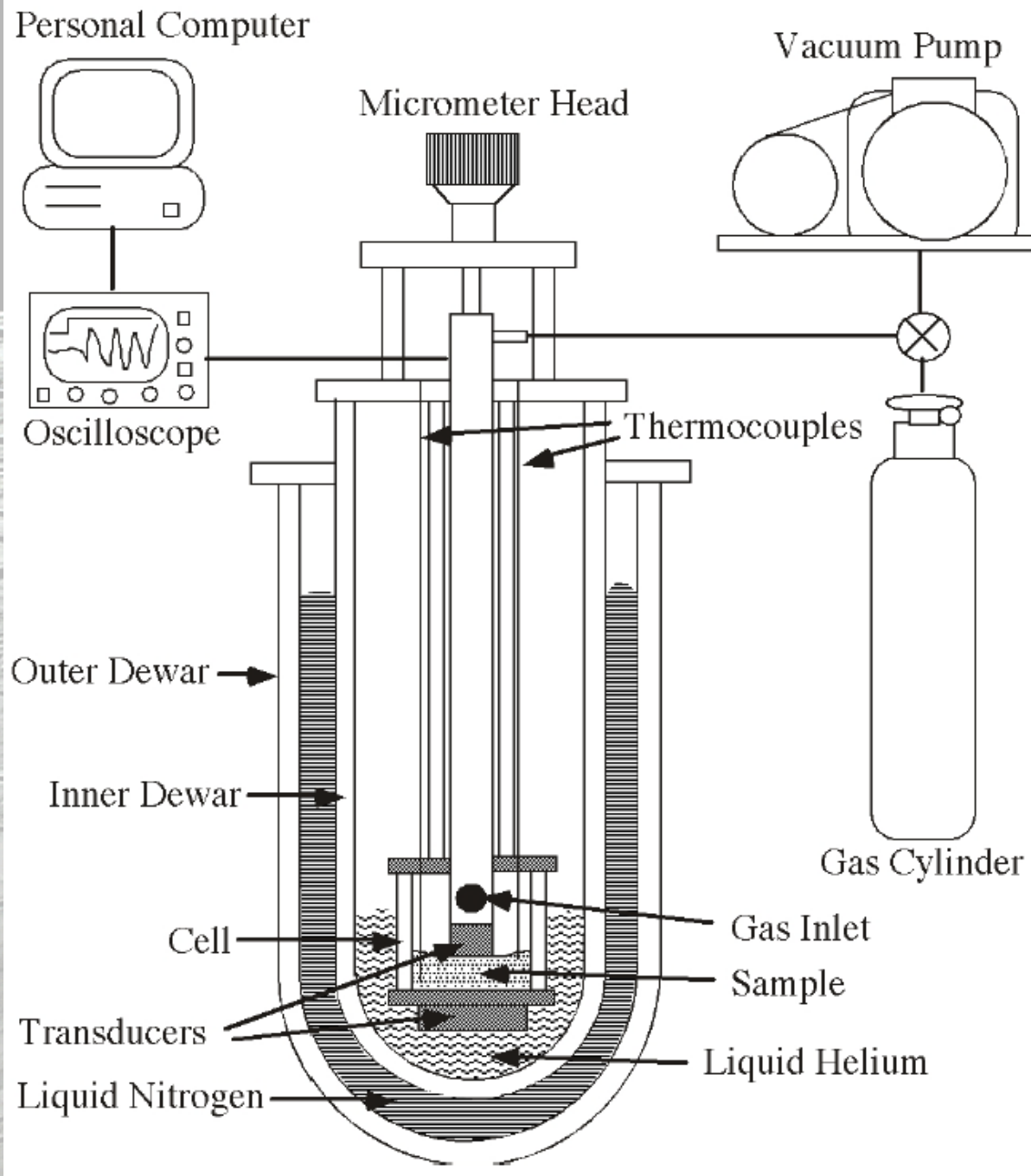
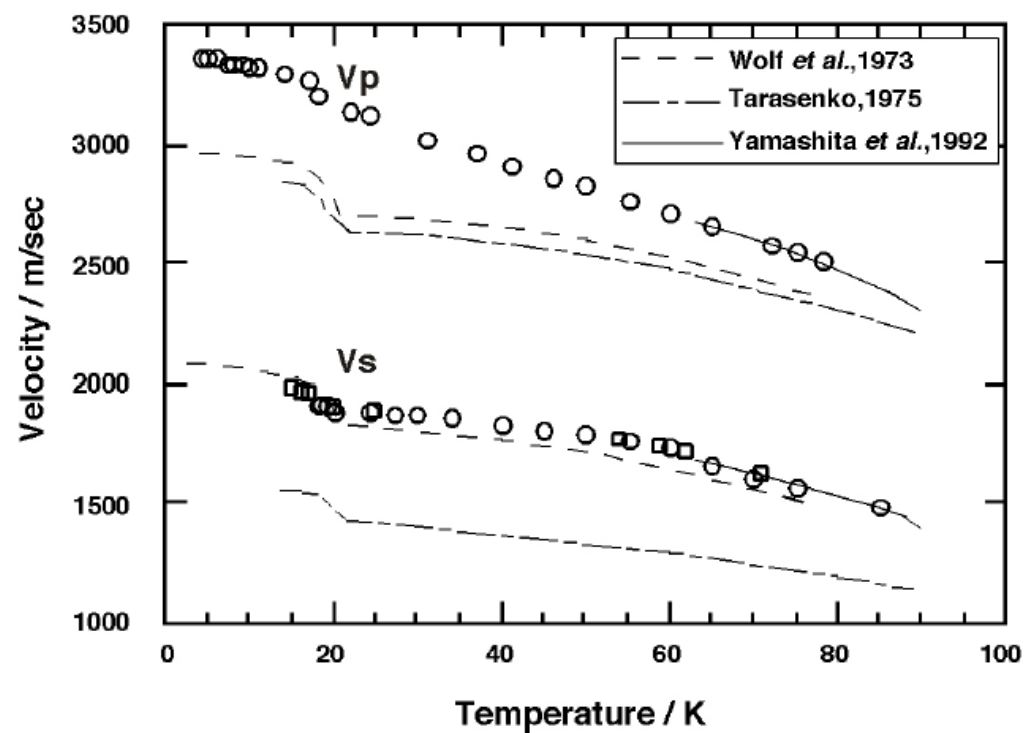
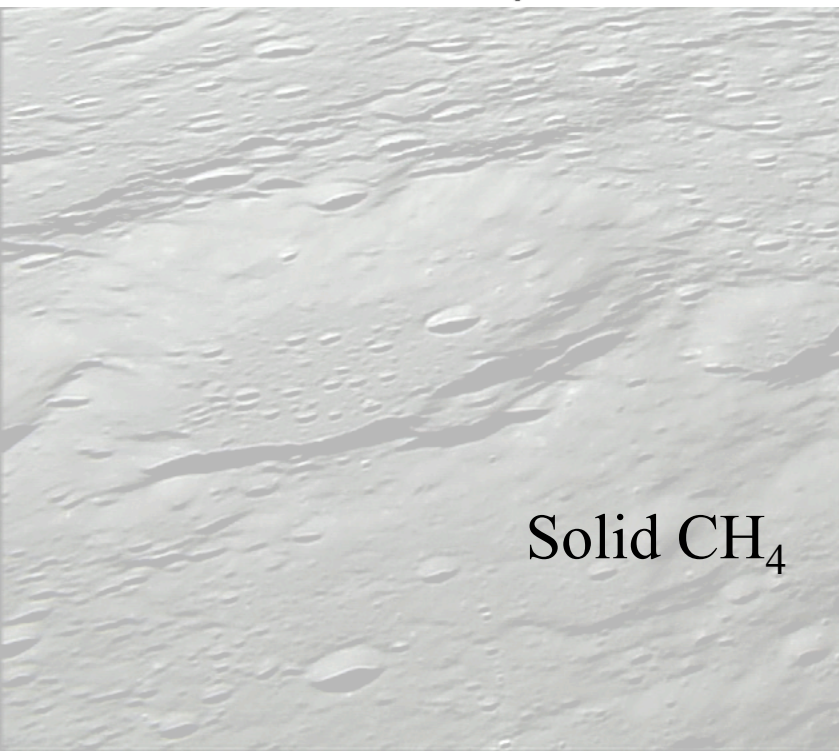
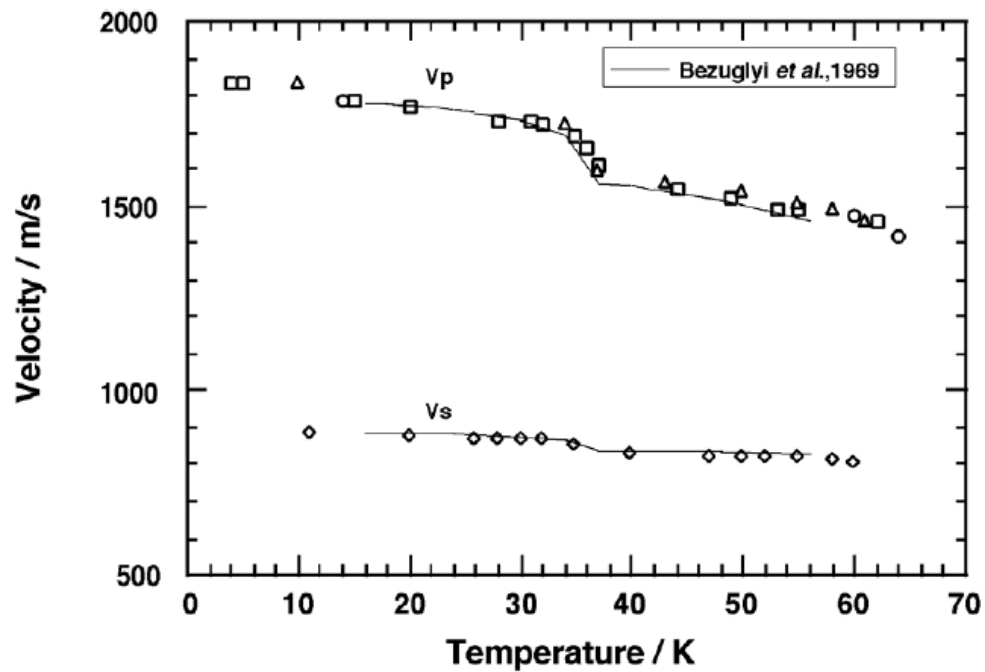


Fig. 1. Compositions, pressures, and phase assignments for $(\text{NH}_3)_x(\text{H}_2\text{O})_{1-x}$ mixtures at 293 ± 3 K, and the isothermal phase diagram for the water-rich region derived from those data. Symbols represent individual data: circle, liquid; cross, Ice VI; dot, Ice VII; triangle, Ice VII + liquid; box, ammonia monohydrate; triangle with cross, Ice VI + liquid; box with cross, monohydrate + Ice VII. solid lines are phase boundaries assigned from these data. Dashed lines are phase boundaries reported by Johnson and Nicol (1987). The precision of the pressure measurement is approximately 0.2 GPa.

Sound Velocity Measurements of N_2 and CH_4 Ices

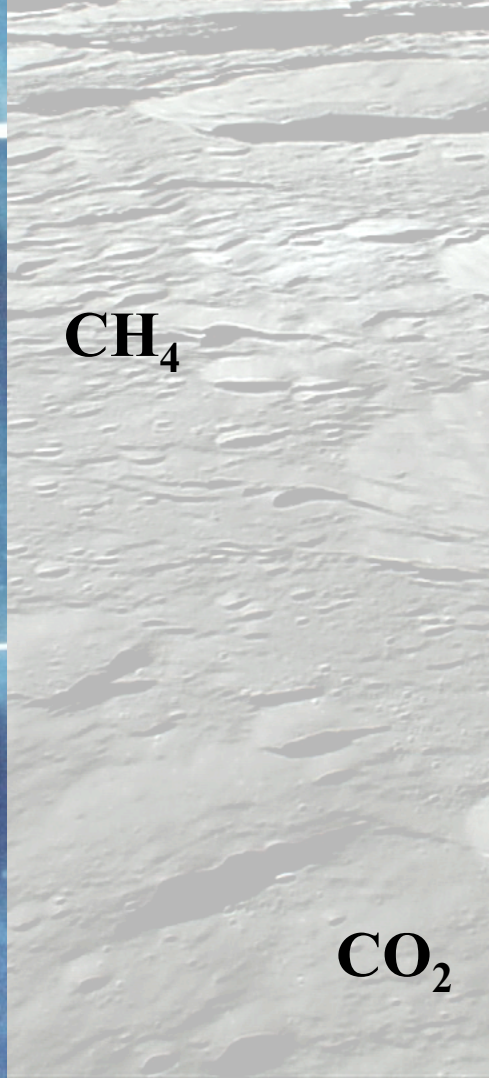
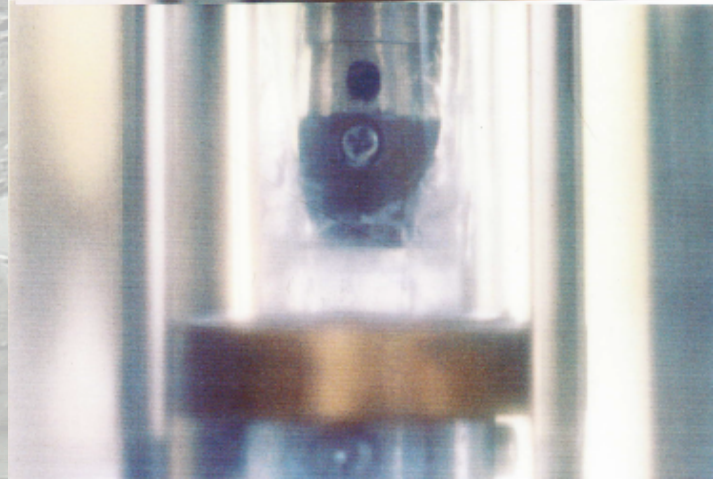
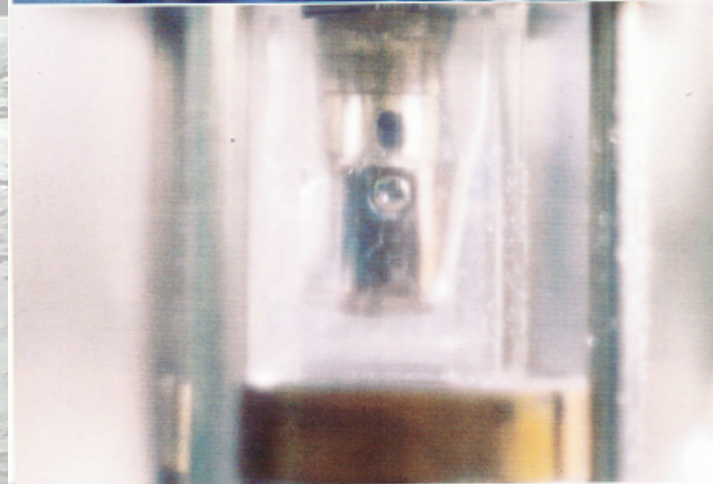
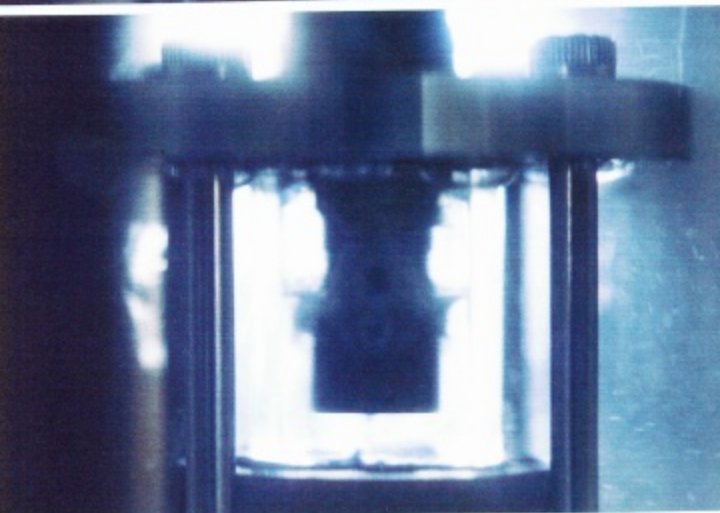
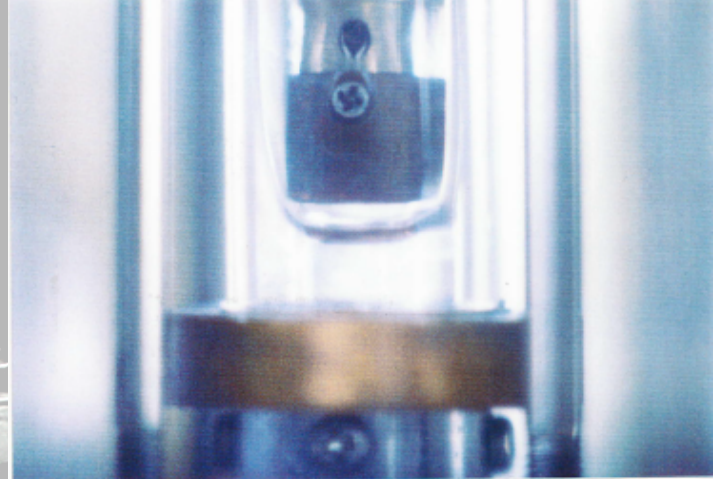
Yamashita et al., Icarus 207, 2010







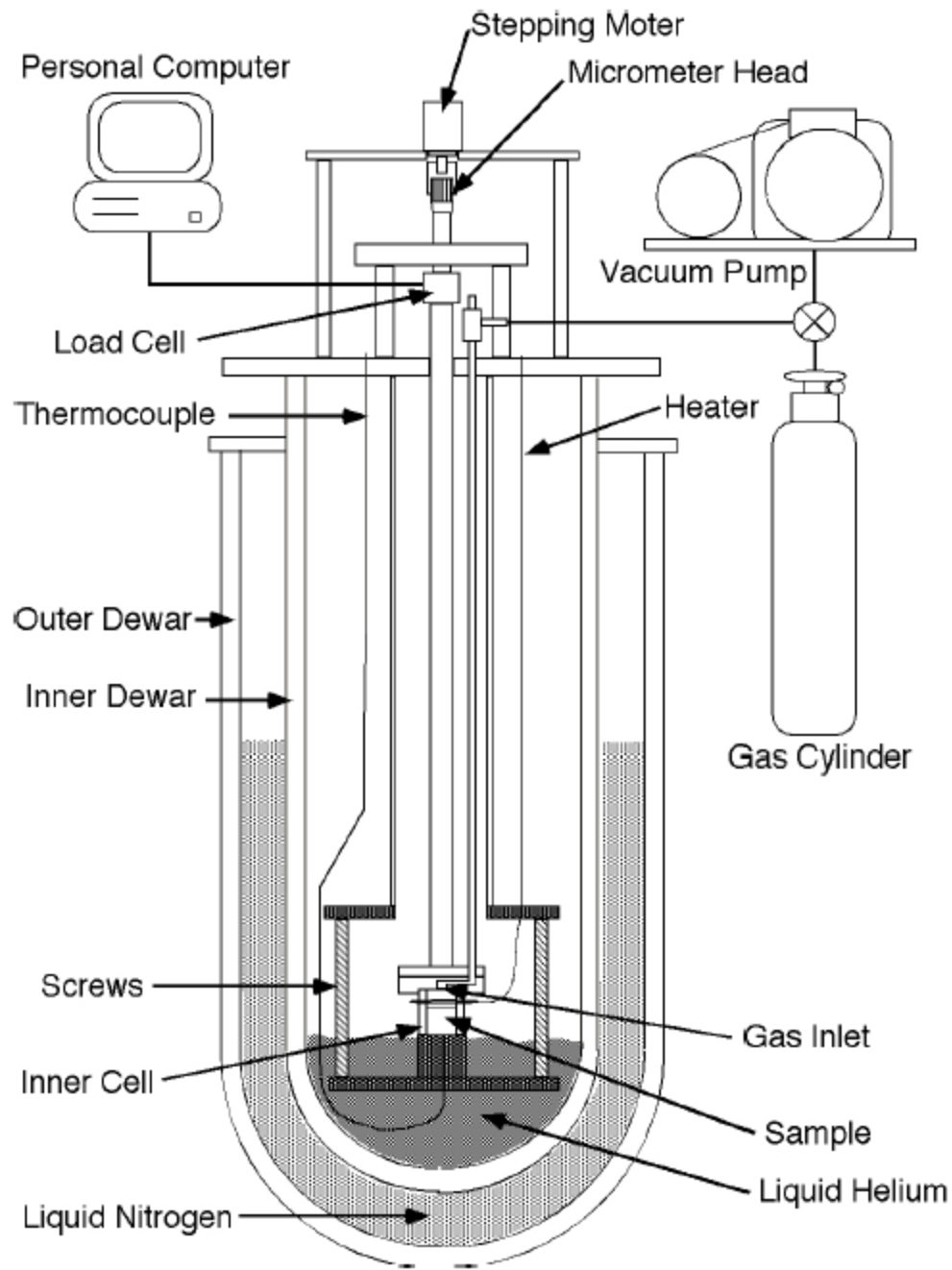
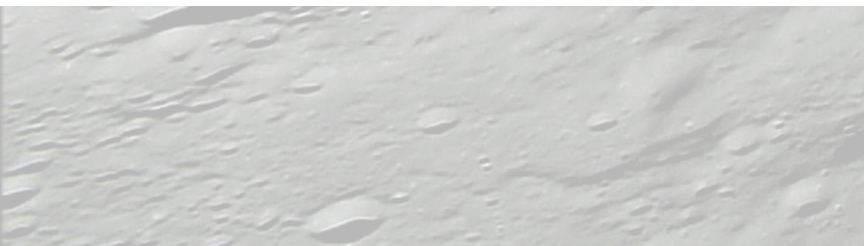
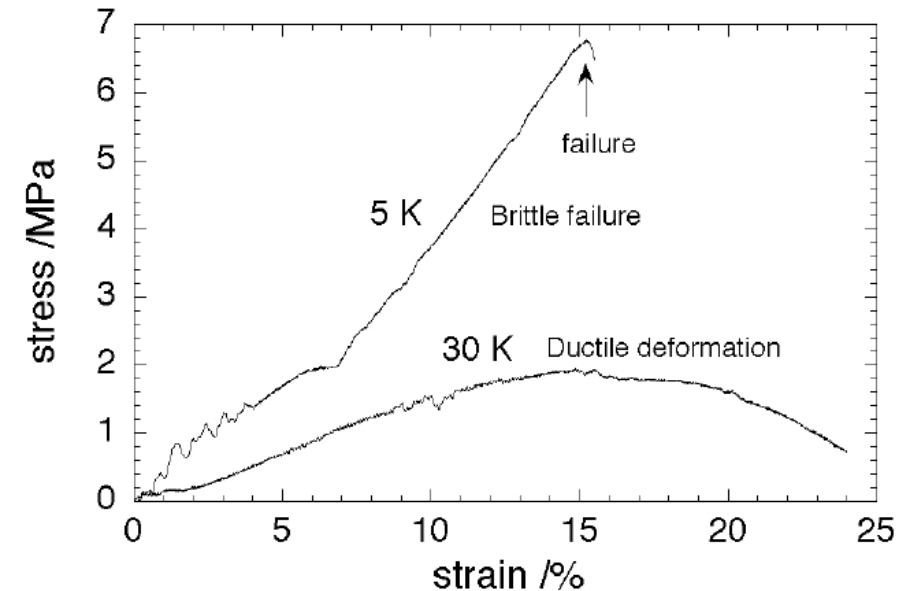
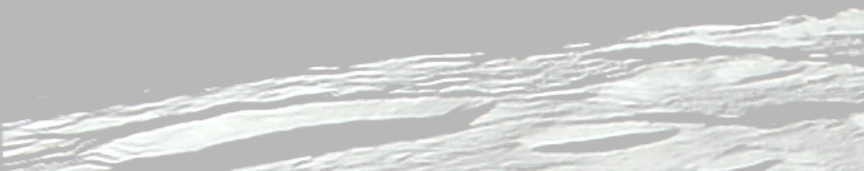
Deformation Experiments At Liquid N₂ Temperature



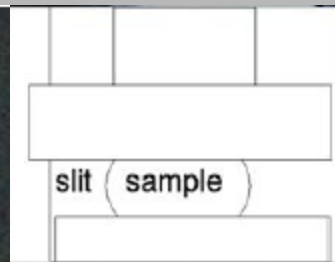
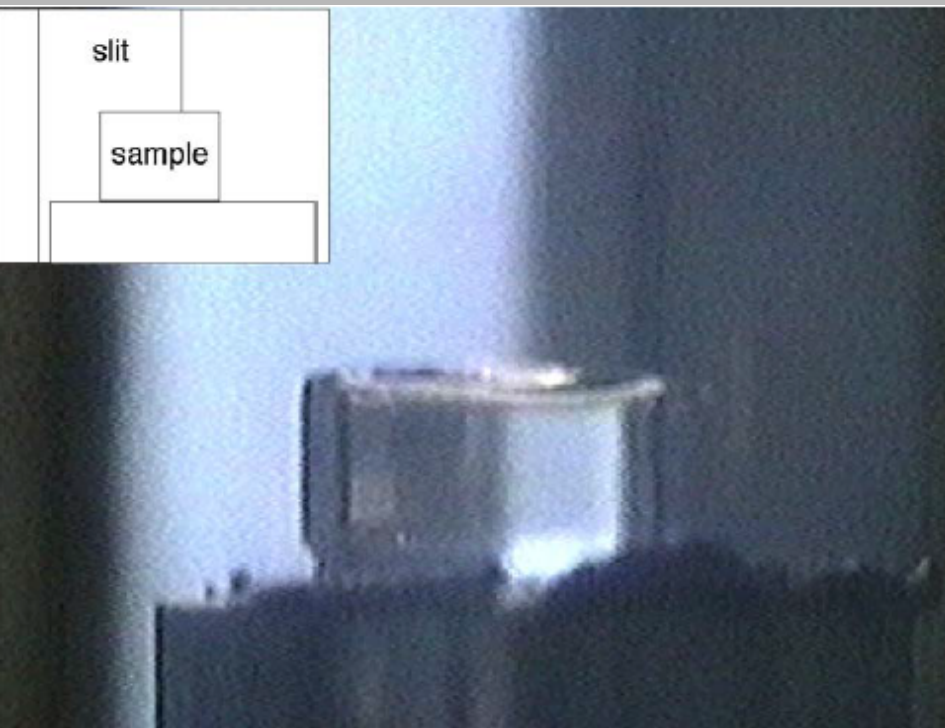
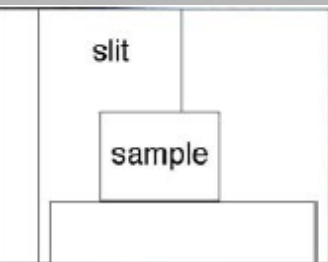
CH₄

CO₂

Uniaxial Deformation of N_2 and CH_4 Ices to 5 K



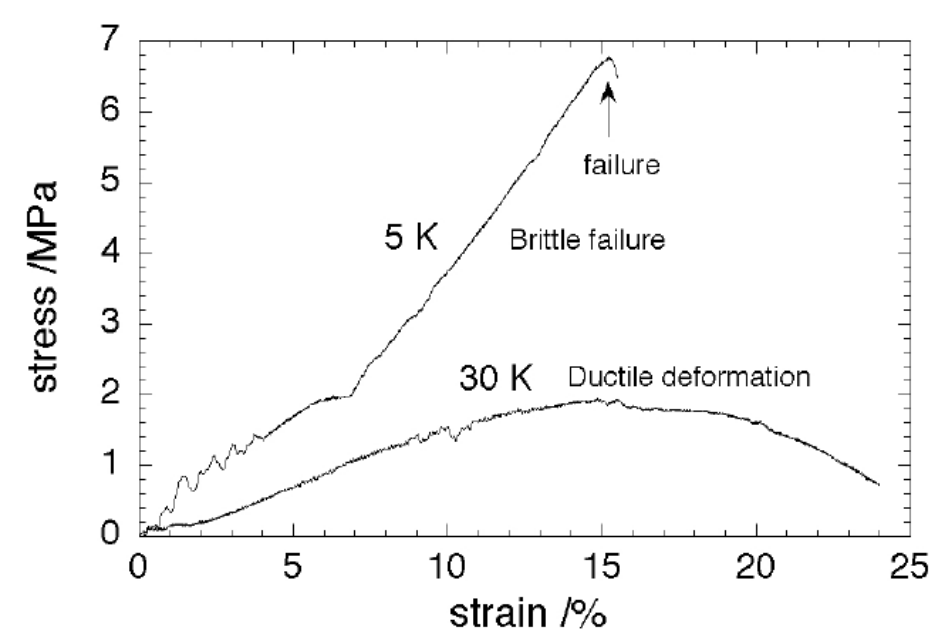
Solid CH₄ at 45 K



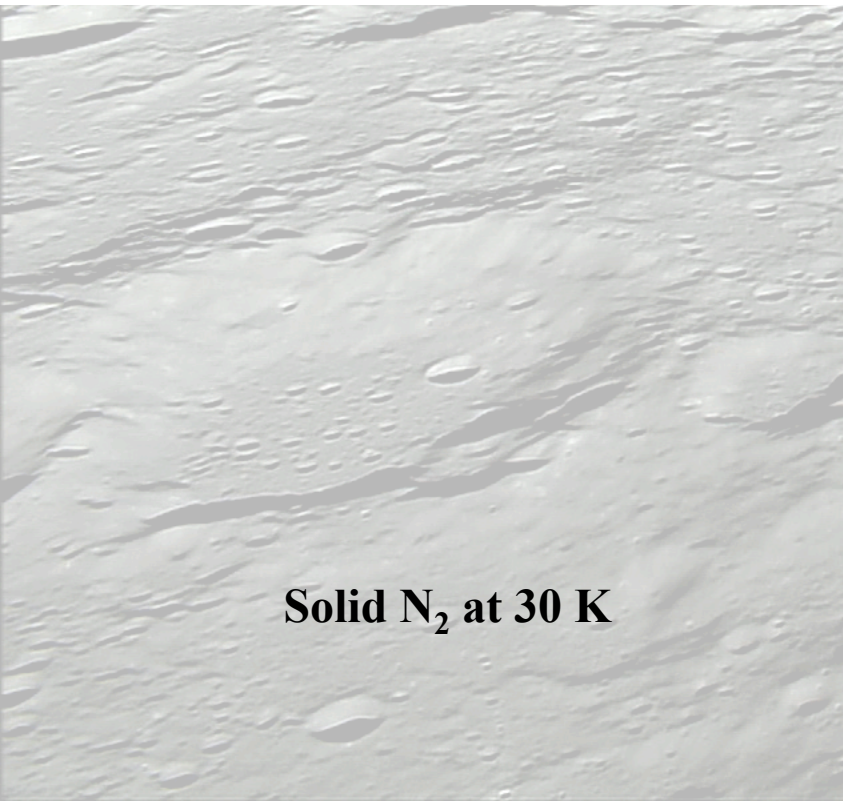
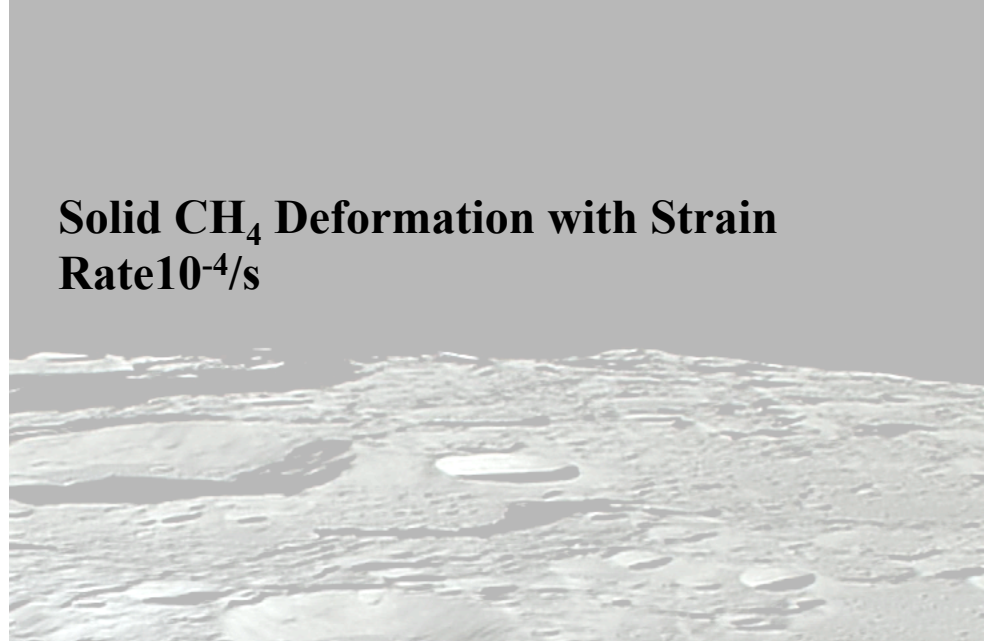
Before Compression

10 mm × 7 mm

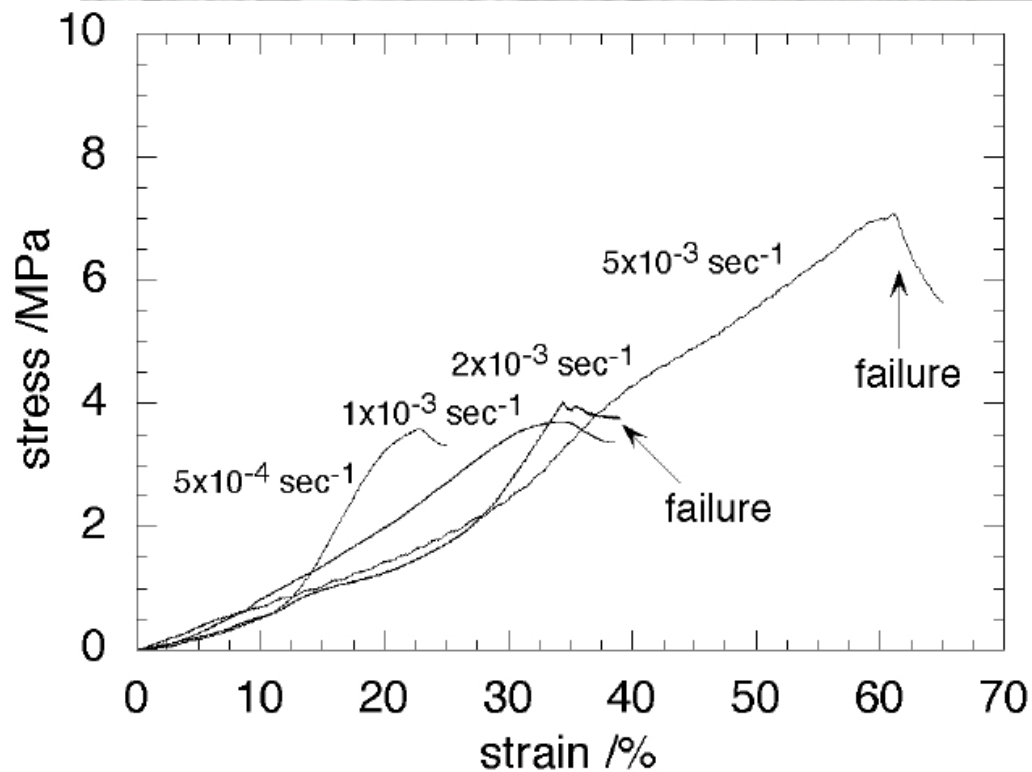
Under Compression
30 % Strain

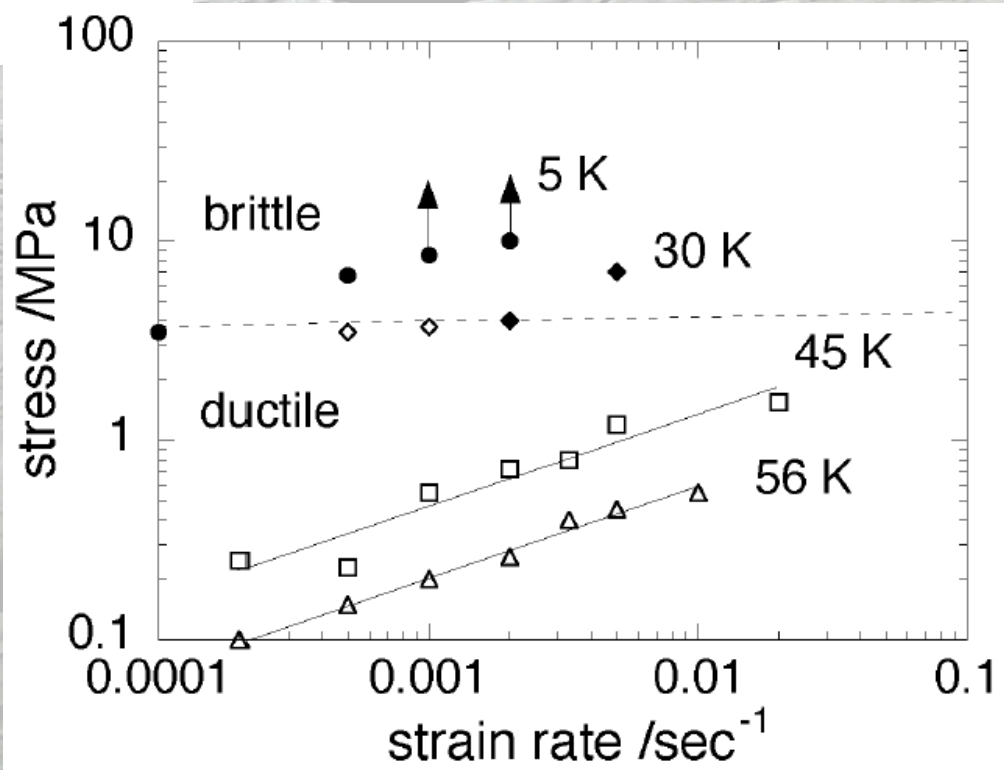
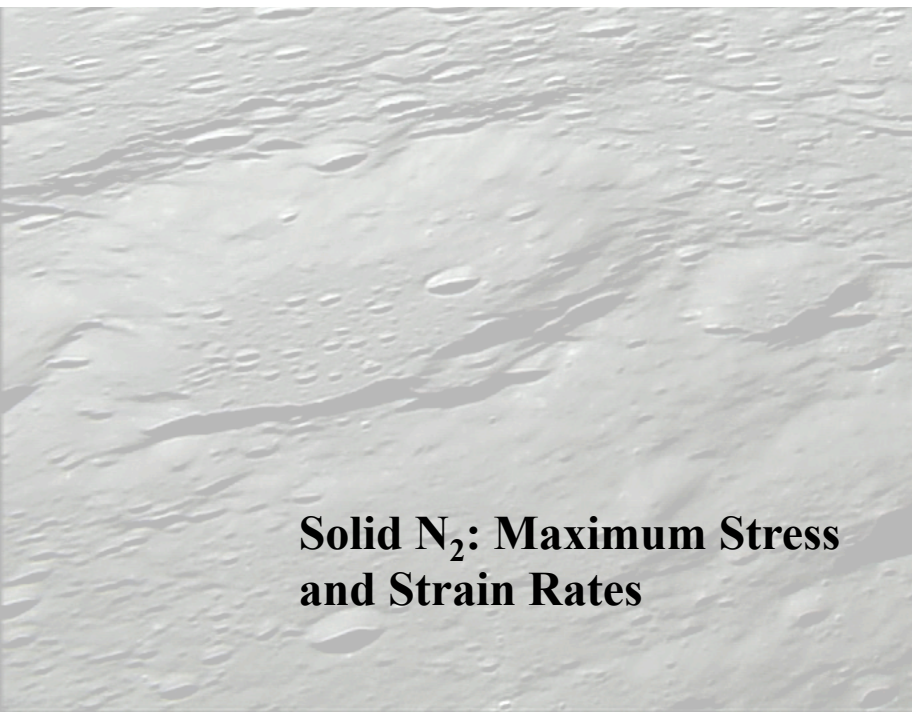
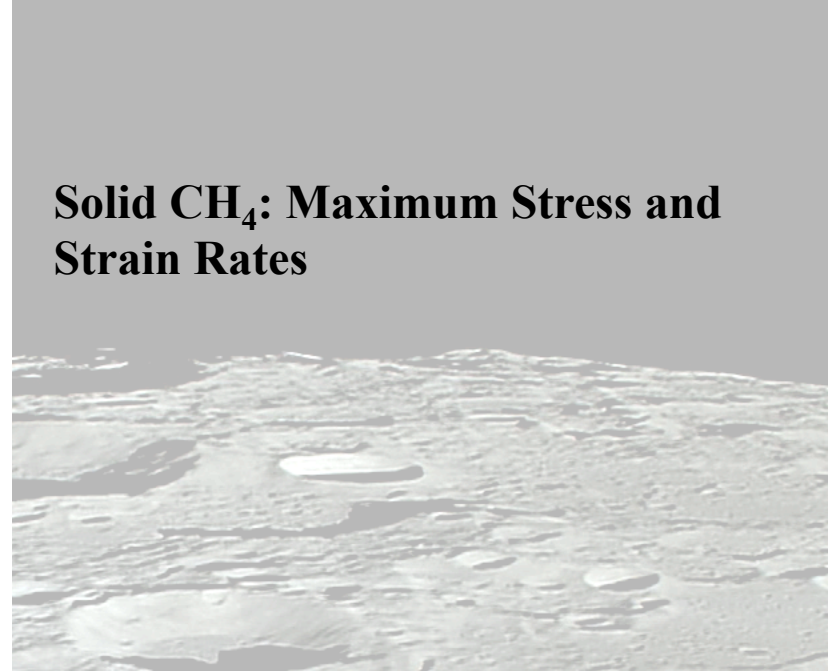
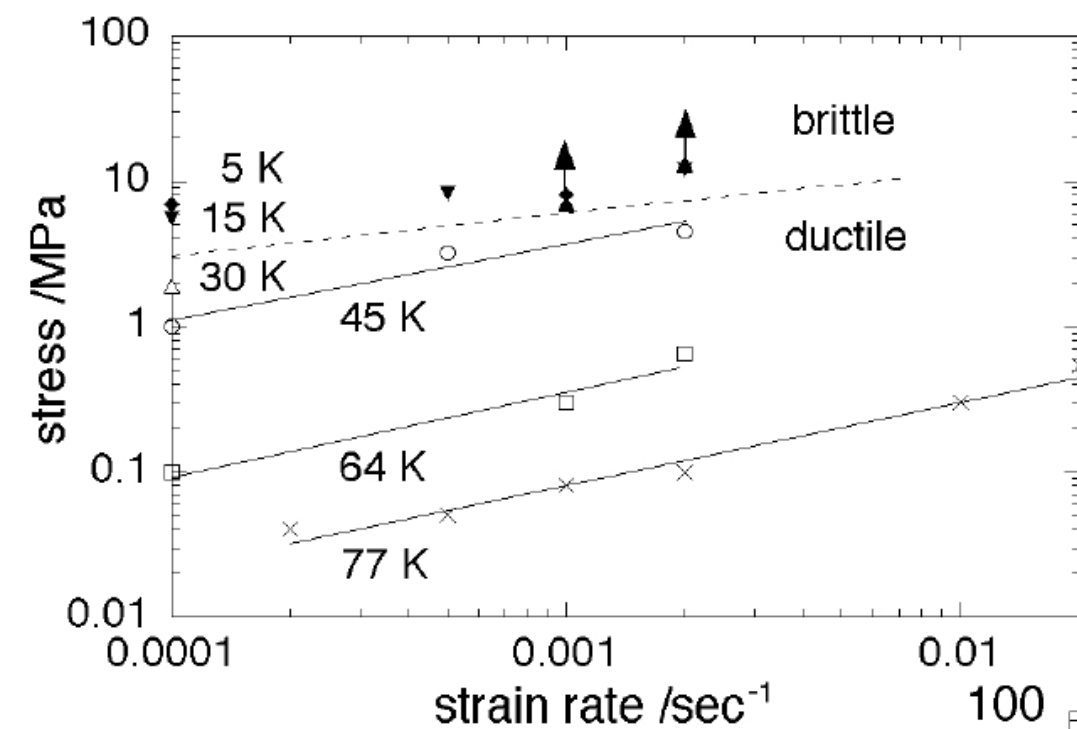


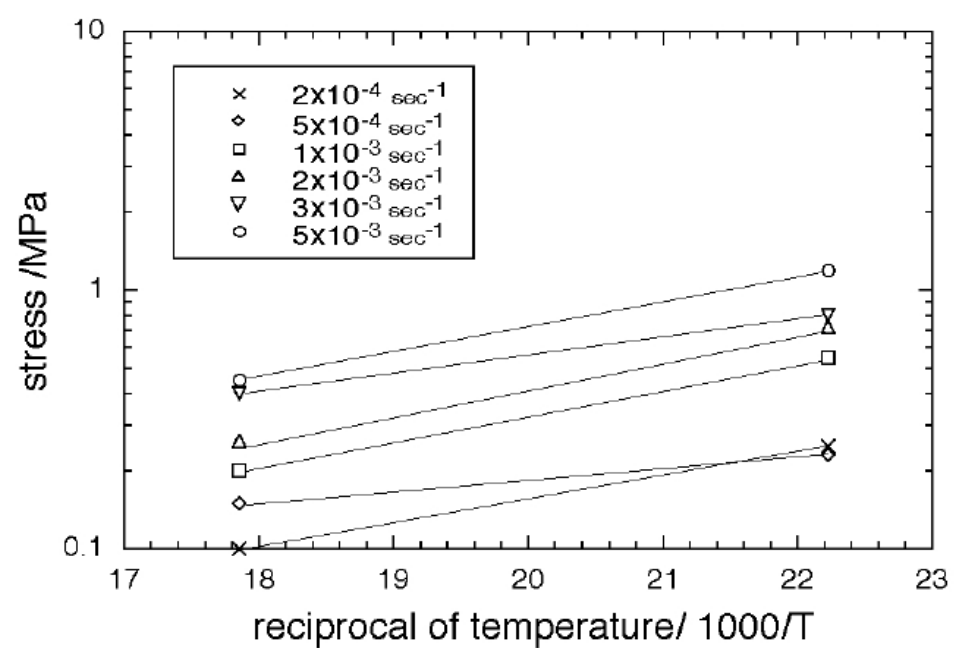
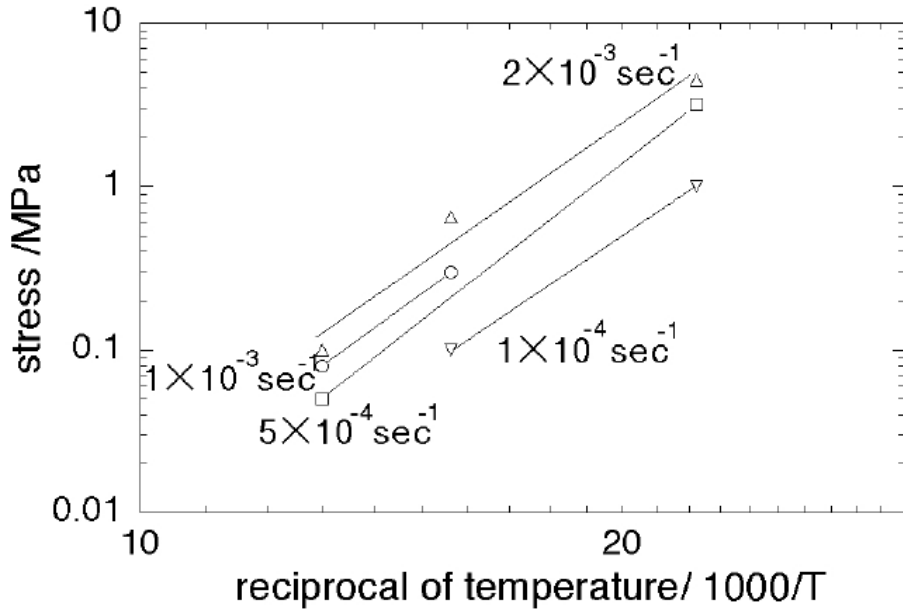
Solid CH₄ Deformation with Strain Rate $10^{-4}/s$



Solid N₂ at 30 K





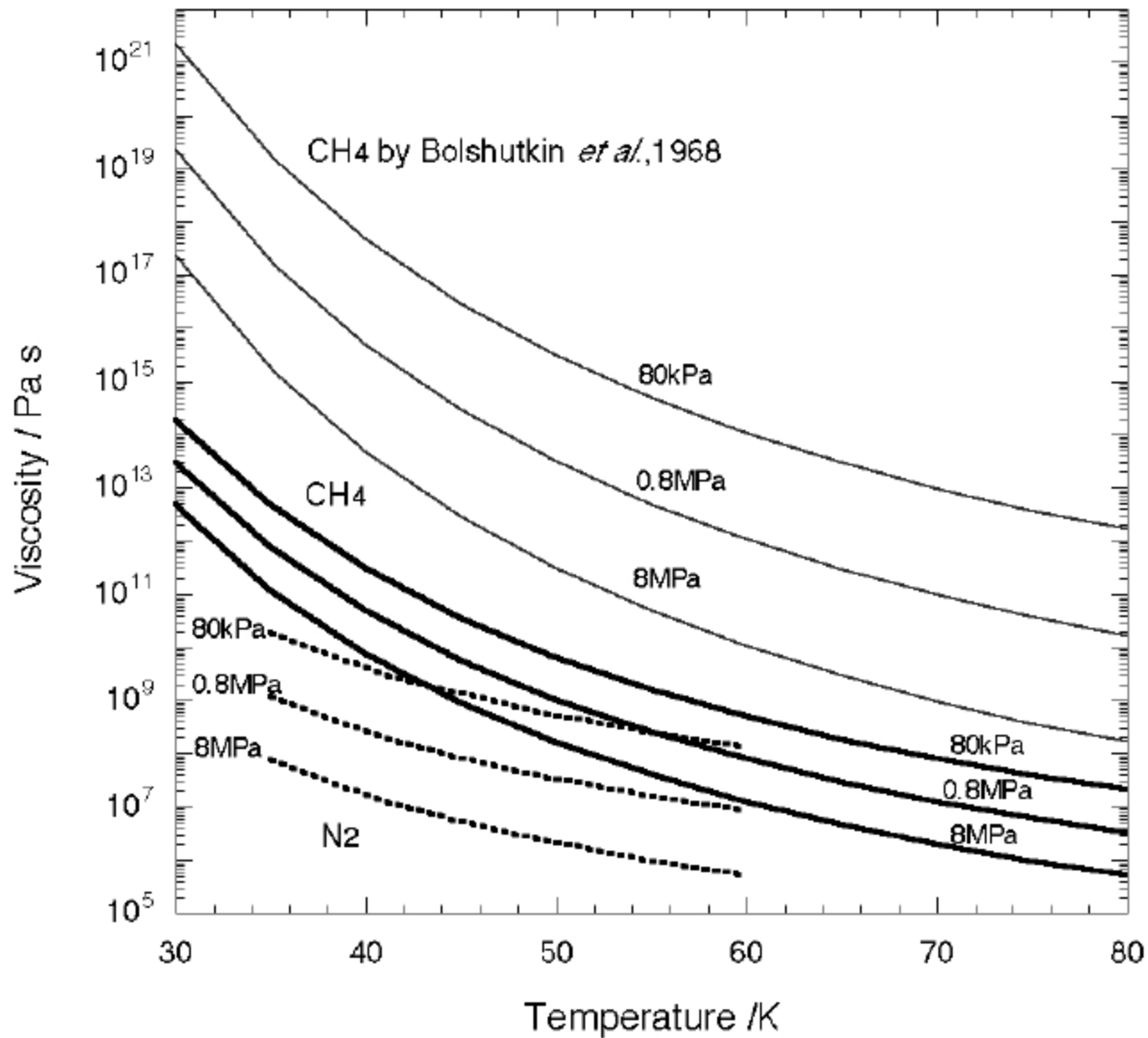


**Solid CH₄: Maximum Stress
and Reciprocal Temperature**

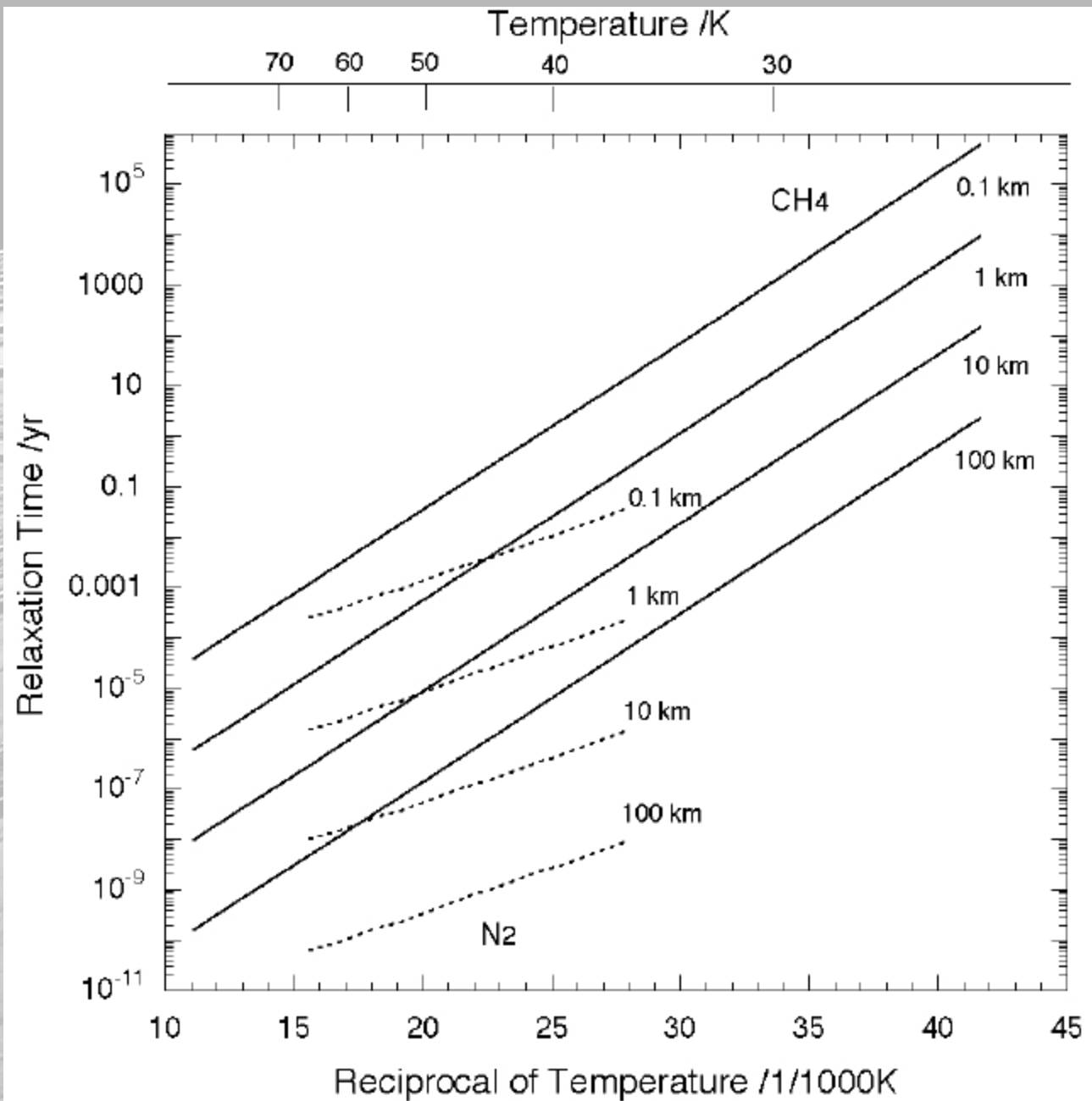
Solid N₂

Flow law parameters for solid nitrogen and methane			
	$\log_{10} A \text{ (MPa}^{-n} \text{s}^{-1}\text{)}$	n	$Q \text{ (kJ/mol)}$
N ₂	3.3 ± 0.4	2.2 ± 0.2	3.5 ± 0.3
CH ₄	1.8 ± 0.4	1.8 ± 0.2	6.4 ± 0.4

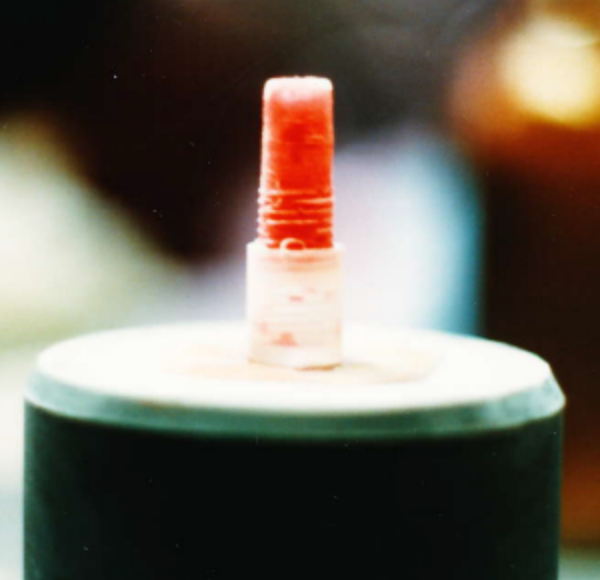
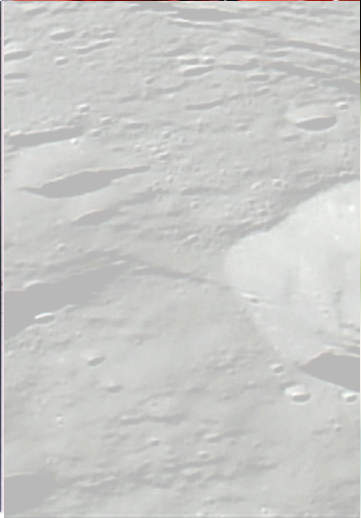
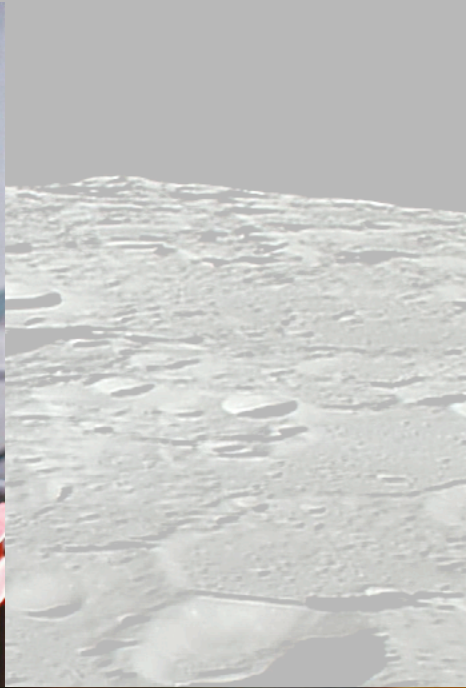
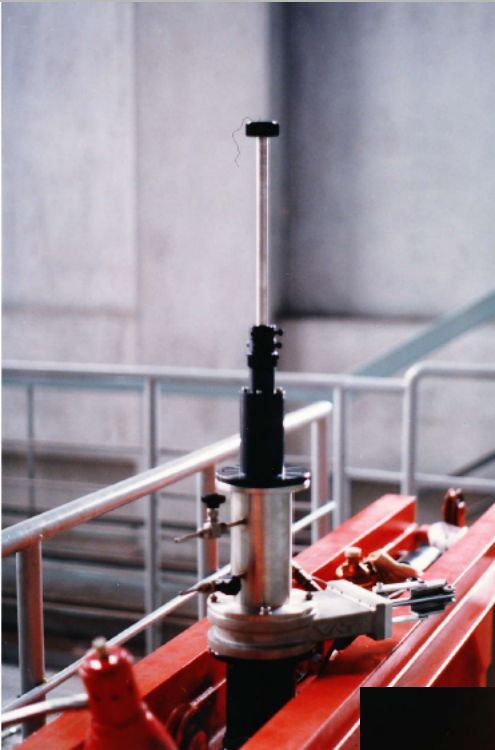
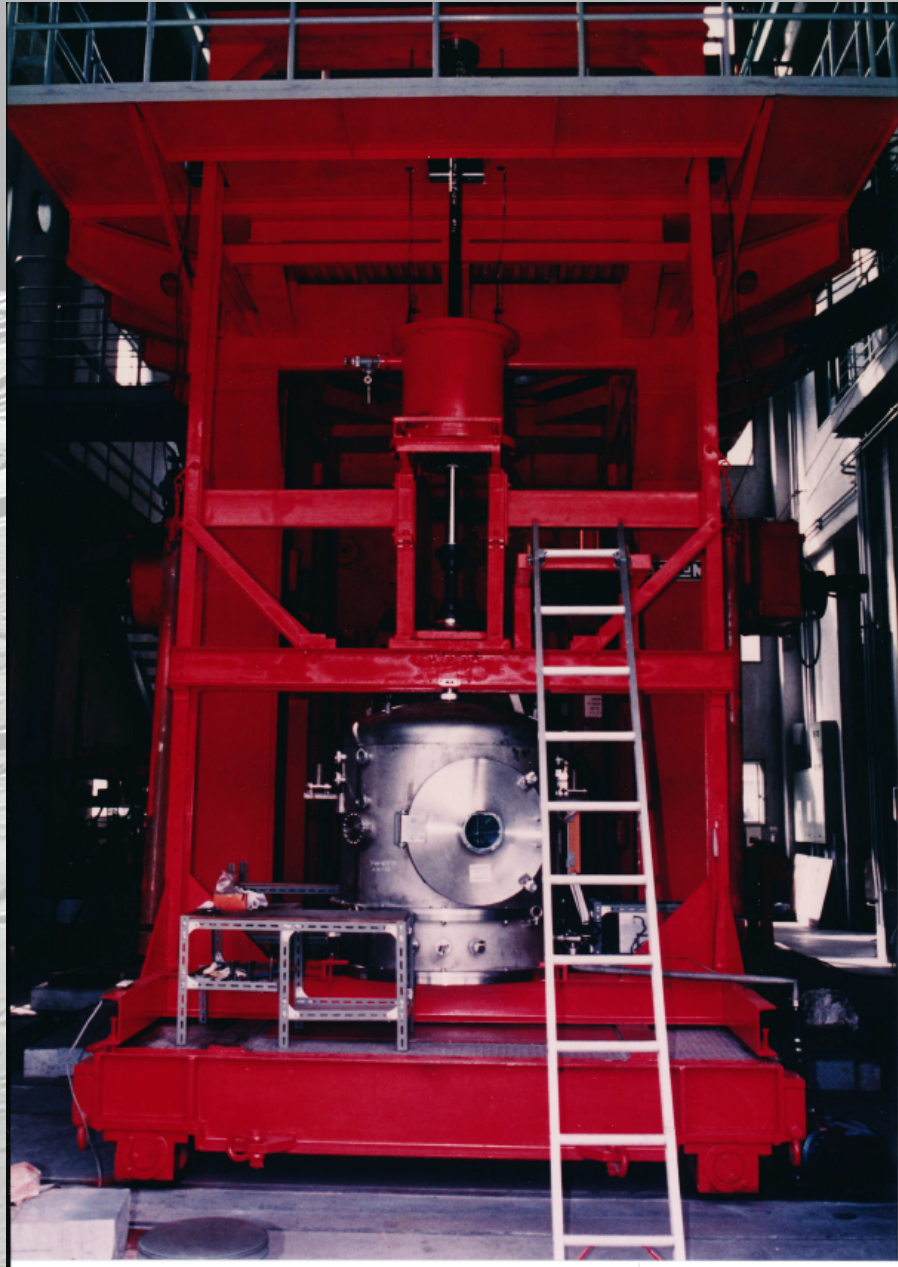
Effective Viscosity and Temperature



Relaxation Times of Craters on Triton Surface



実験惑星学4: Ice Impact Experiments



Institute of Low Temperature Science

