

**EXPERIMENTAL MID-INFRARED
SPECTROSCOPIC ANALYSIS:
ITS APPLICATION AND POSSIBILITY**
室内実験における赤外分光分析の利用法と可能性

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CPS, Kobe Japan**

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Outline

- Introduction
 - IR spectroscopy
- IR spectroscopic measurements
 - IR spectroscopy
 - * Function
 - * Aerosol spectroscopy
 - * In situ FTIR spectroscopy
 - * IR ellipsometry
 - Sample preparation
 - * Pellet
 - * Aerosol spectroscopy
 - * On substrate
- Measurements & data analysis
- Astronomical application
- Possibilities for future
- Summary

Introduction

◆ What is the IR spectroscopy?

An analytical technique to identify substances



Quartz
(SiO_2)



Corundum
(Ruby: Al_2O_3)



Olivine
($\text{Mg}_{1.9}\text{Fe}_{0.1}\text{SiO}_4$)



Kyanite
(Al_2SiO_5)



Aquamarine
($\text{Be}_3\text{Al}_2(\text{SiO}_3)_6$)



Enstatite
(MgSiO_3)



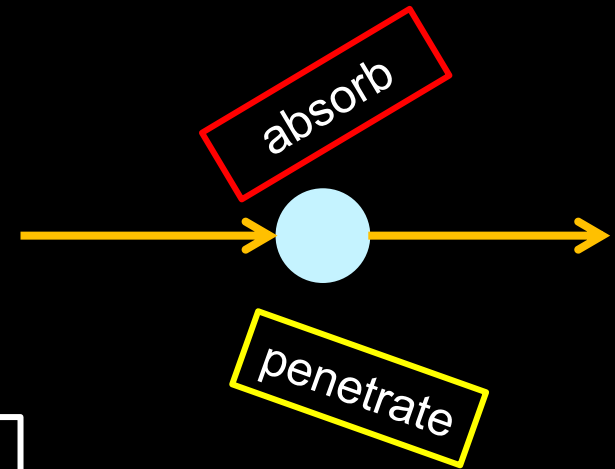
Amo- MgSiO_3



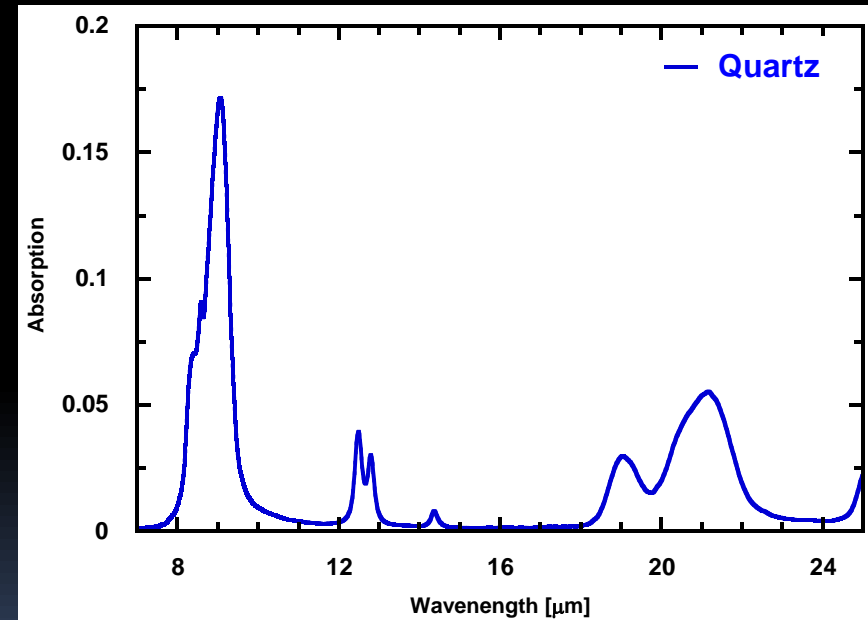
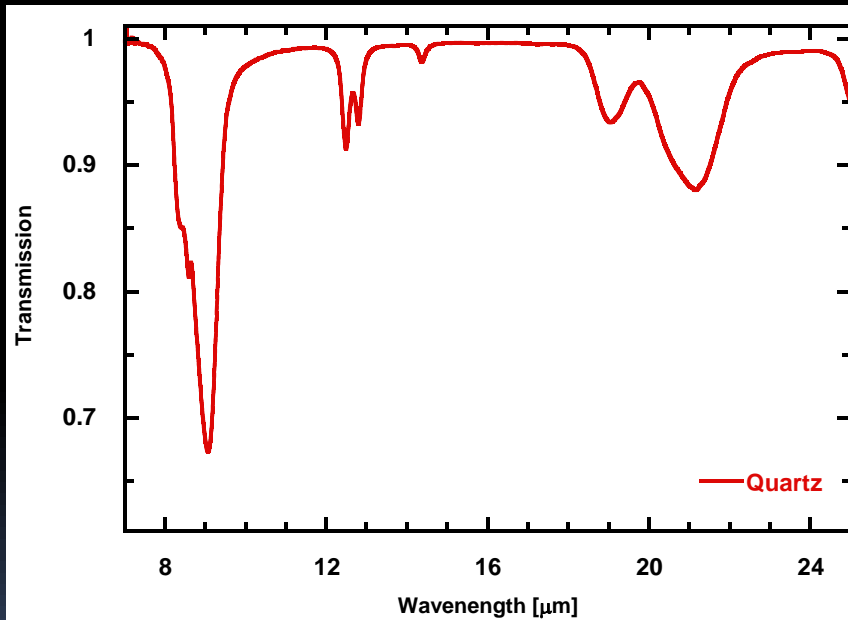
Fayalite
(Fe_2SiO_4)

What can we obtain?

Detecting absorption and transmission bands due to vibrational transitions of atomic bonds



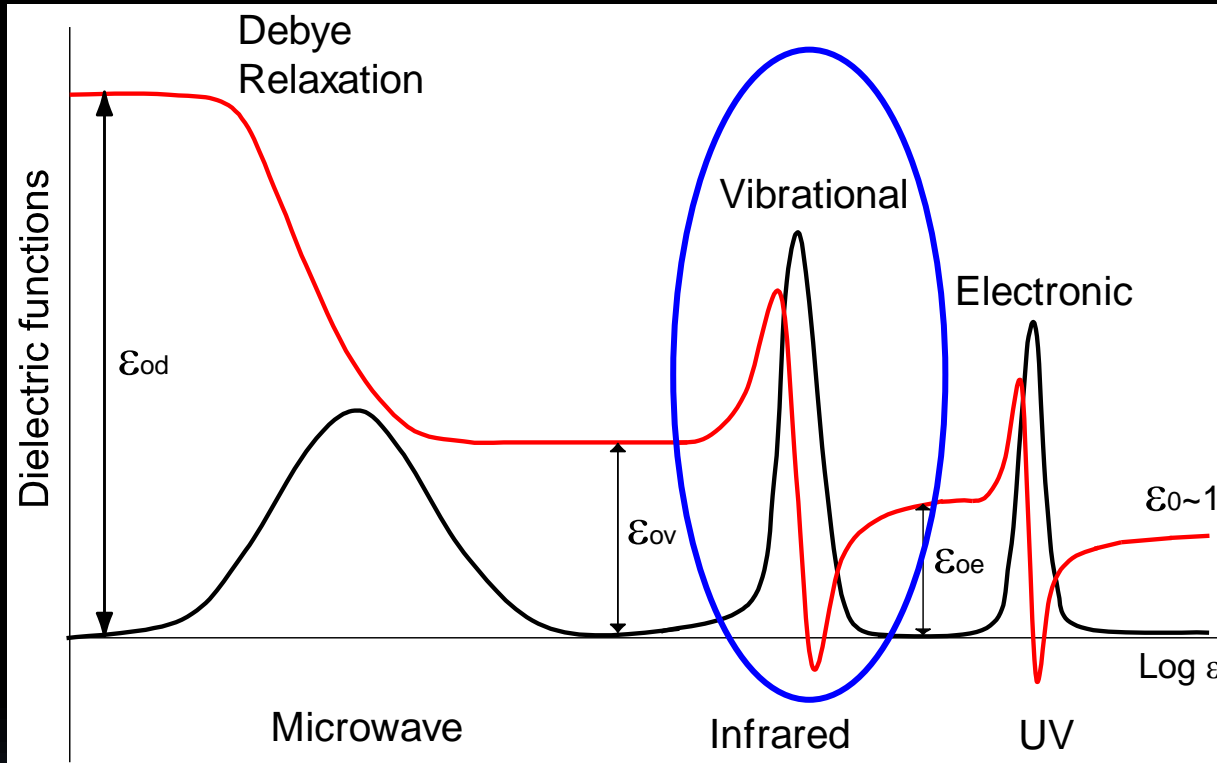
Crystalline SiO₂



Absorbance = $-\log(\text{transmittance})$
Transmittance = $10^{-\text{absorbance}}$

Fingerprint of the material along with absorption peaks

Optical constant & Absorption



Describing optical properties

$$m = n + ik$$

n : real part
directly relate to refraction
 k : imaginary part
relate to absorption

In microscopic mechanisms \rightarrow Dielectric function

$$\epsilon = \epsilon' + i\epsilon'' = (n + ik)^2$$

$$\epsilon' = n^2 - k^2$$

$$\epsilon'' = 2nk$$

Electronic transitions inside atoms and molecules
 \rightarrow UV regions

Rotational spectra \rightarrow far-IR & microwave regions

Vibrational spectra \rightarrow near- & mid-IR

How can we do the IR spectroscopic measurements?

IR spectroscopy

- FTIR spectrometer
- Aerosol spectroscopy
- In-situ FTIR spectroscopy
- IR ellipsometry

Sample preparation

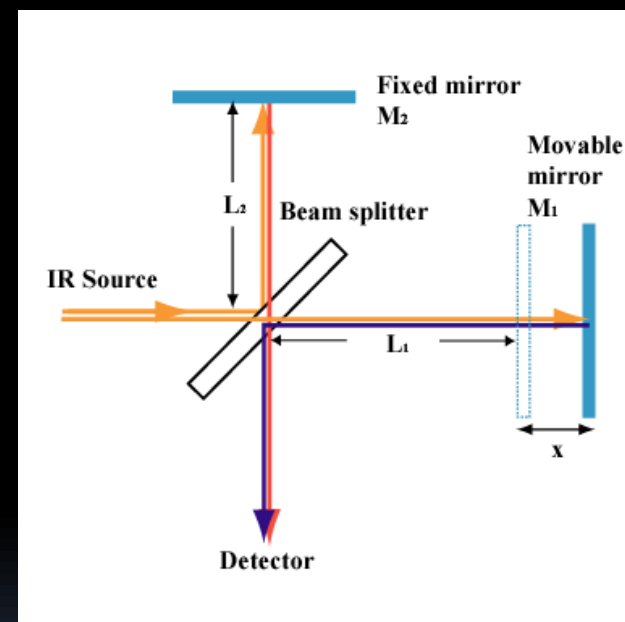
- Pellet technique
- Aerosol technique
- On substrate
 - evaporation
 - spin coating

FTIR Spectrometer

Fourier Transformation Infrared Spectroscopy (FTIR)

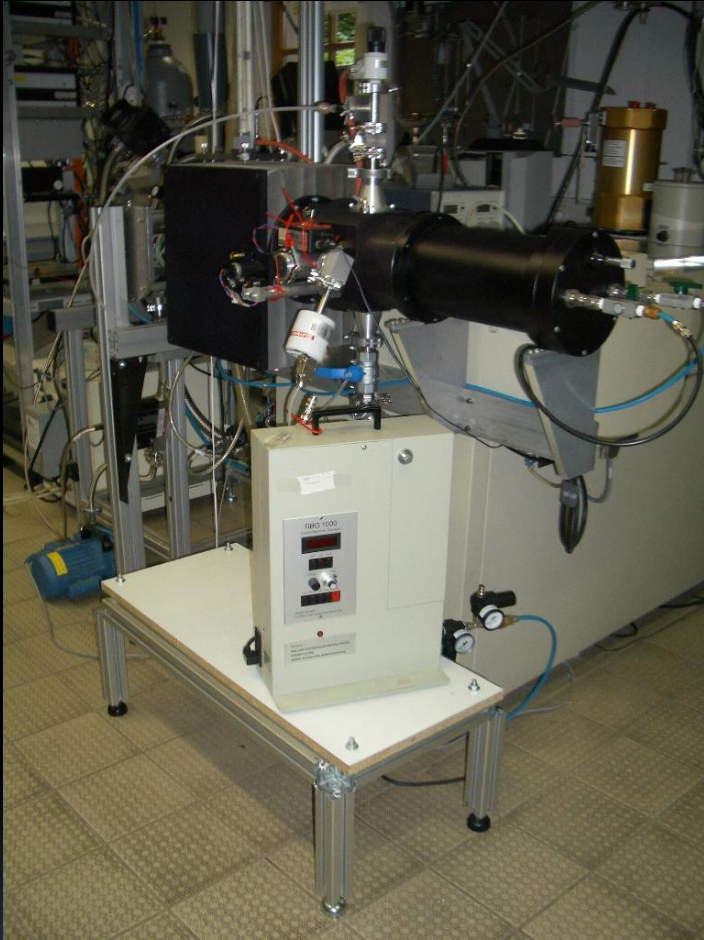
Michelson interferometer → the heart of FTIR & using for encoding a composite signal that contains all of the IR frequencies present in order to measure all of the IR frequencies at the same time

- ✓ A beam-splitter (w/ an orientation angle of 45° relative to the 2 mirrors) to divide an incident beam into 2 beams
- ✓ A fixed mirror is located at \parallel to the incident beam w/ a distance L_2 from the splitter
- ✓ A movable flat mirror is placed at \perp to the beam with a distance (L_1+x) from the splitter
- ✓ $\frac{1}{2}$ of the beam reflects off the splitter and travels to the fixed mirror M_2
- ✓ The other half of the beam passes through the beam splitter and reflects off the movable mirror M_1 precisely moves back and forth along the direction of the incident beam within a distance x
- ✓ After both beams reflect off the mirrors respectively, they recombine at the beam splitter

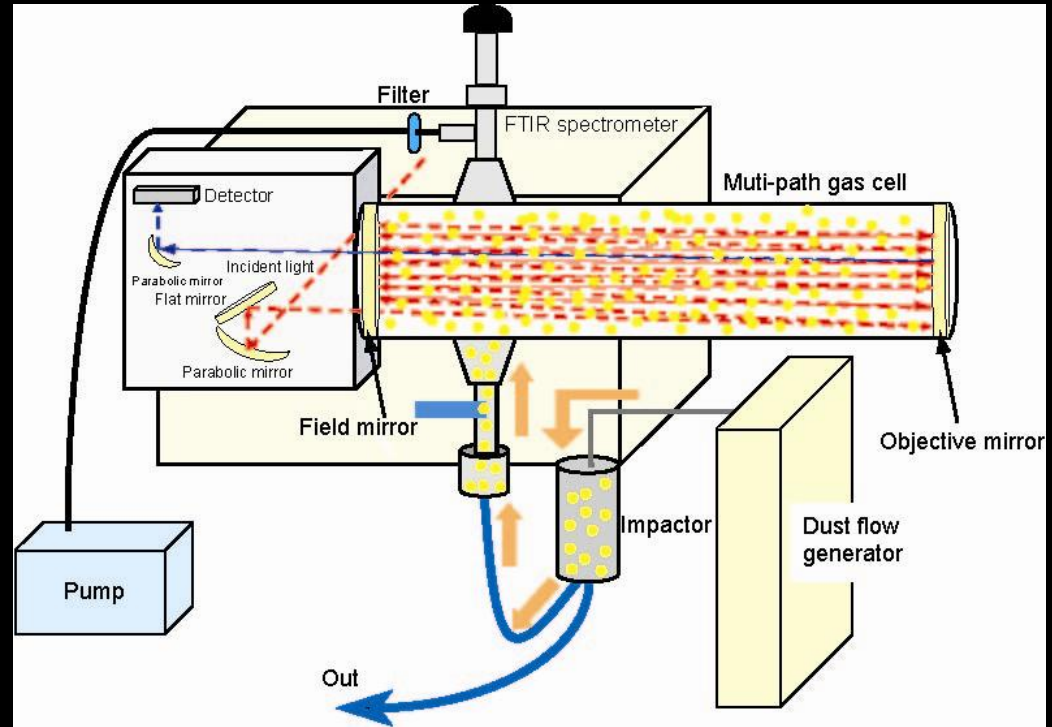


- ✓ As a result of a phase difference between these 2 beams, they interfere w/ each other either constructively or destructively depending on the additional path difference

Aerosol spectroscopy

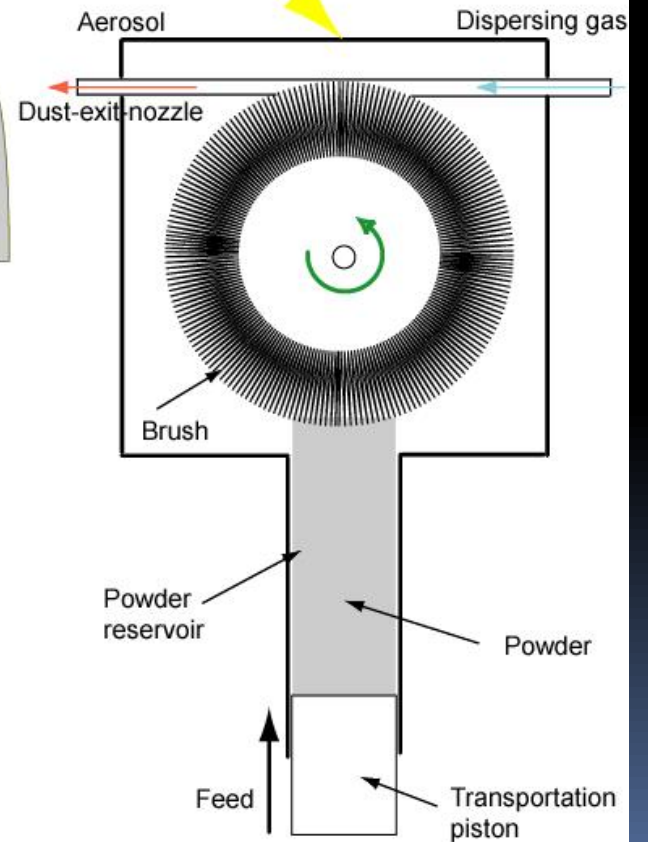
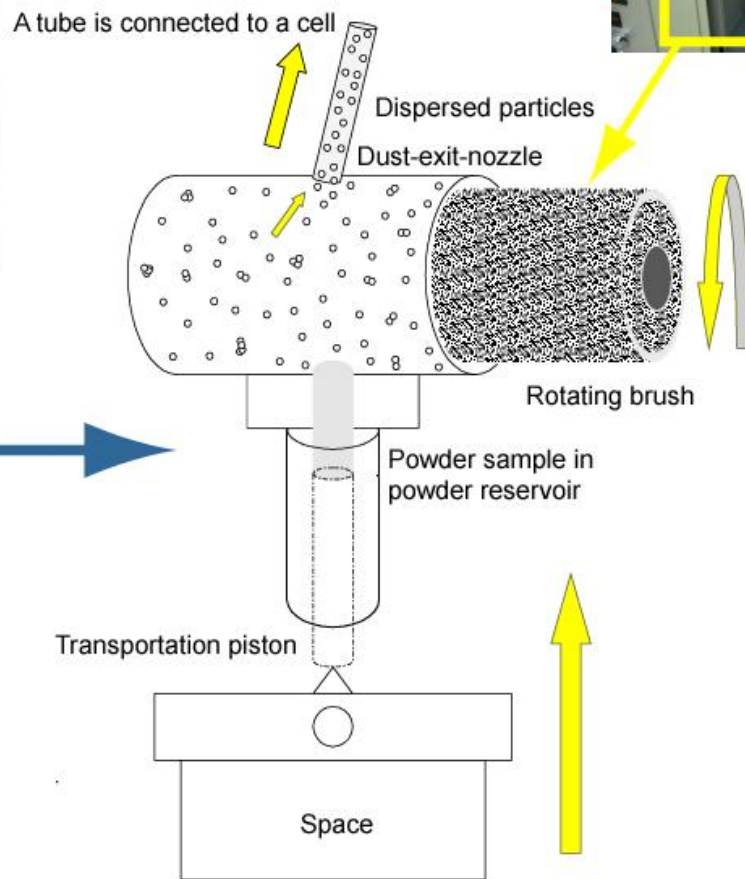
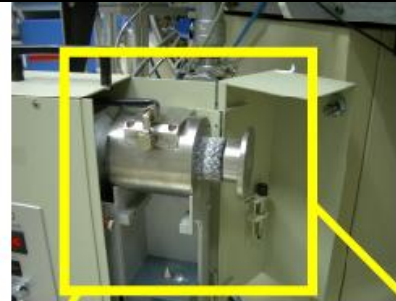


AIU Jena
More details in Fundamentals and
Applications in Aerosol Spectroscopy,
Tamanai & Mutschke 2010

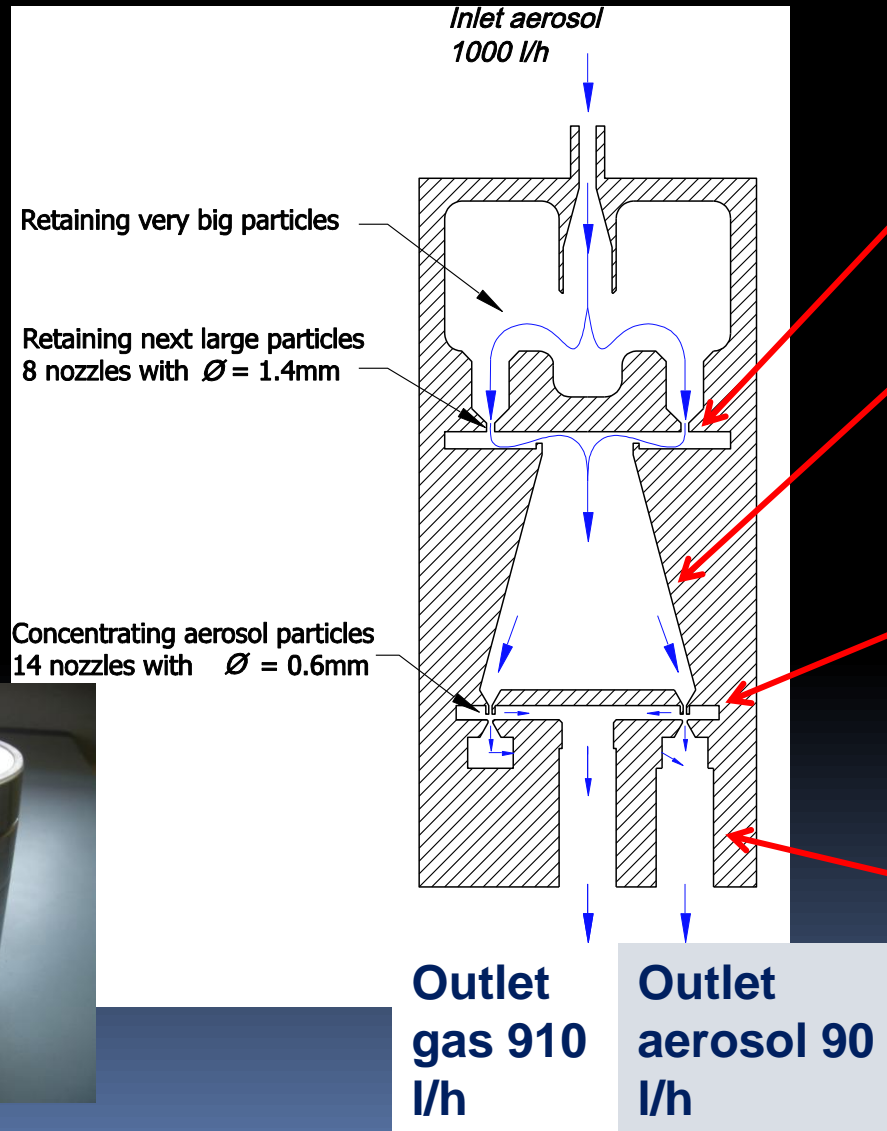


- Particle size: $<1\ \mu\text{m}$
- N_2 -gas
- Path length: $\sim 18\text{m}$
- Wavelength range: $2\text{-}50\ \mu\text{m}$
- Sampling:
Polyester capillary pore
membrane filters

Aerosol Generator

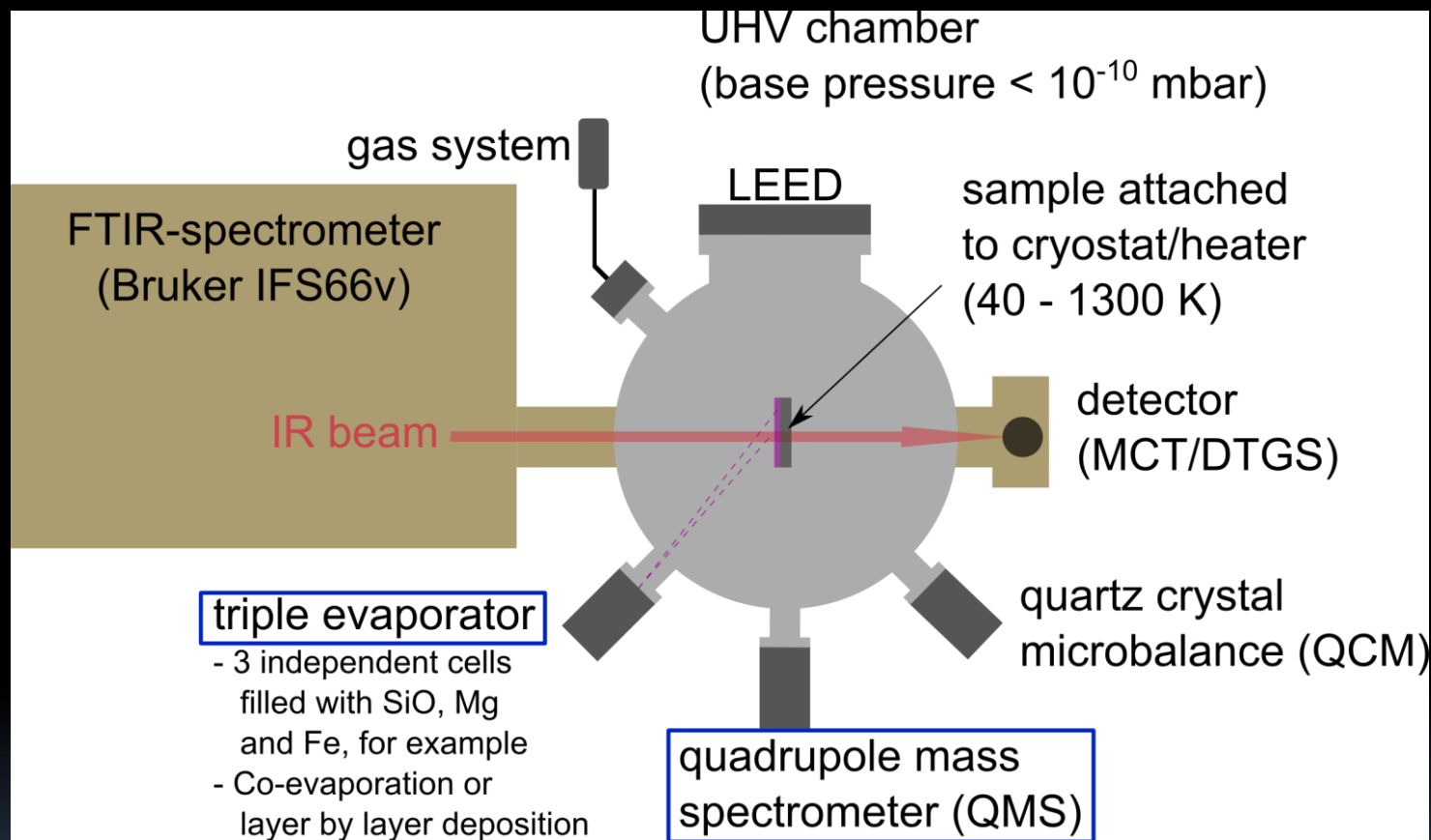


Impactor



In situ FTIR spectroscopy

Condensation experiment



Triple evaporator for preparation of different types of thin films:



IR Ellipsometry

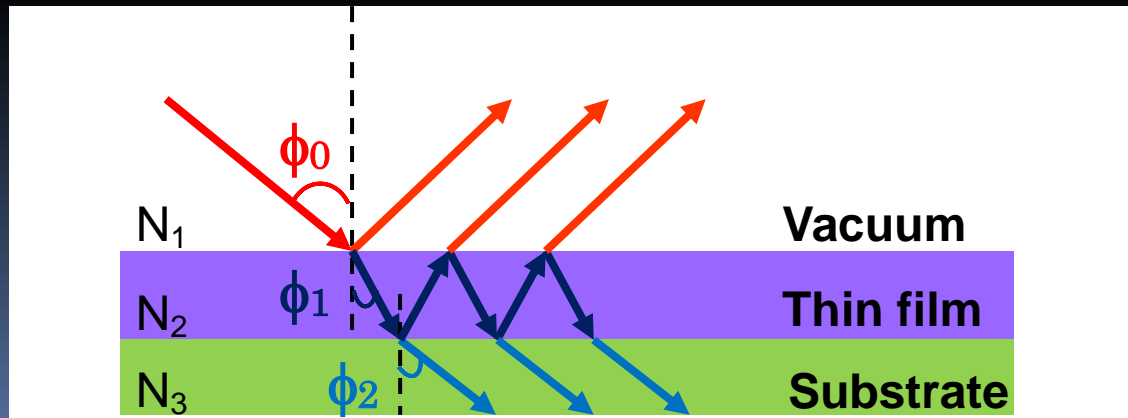
Ellipsometry measurements?

- Possible to investigate the optical and physical properties of thin films and bulk materials
 - ❖ Film thickness
 - ❖ Dielectric constant

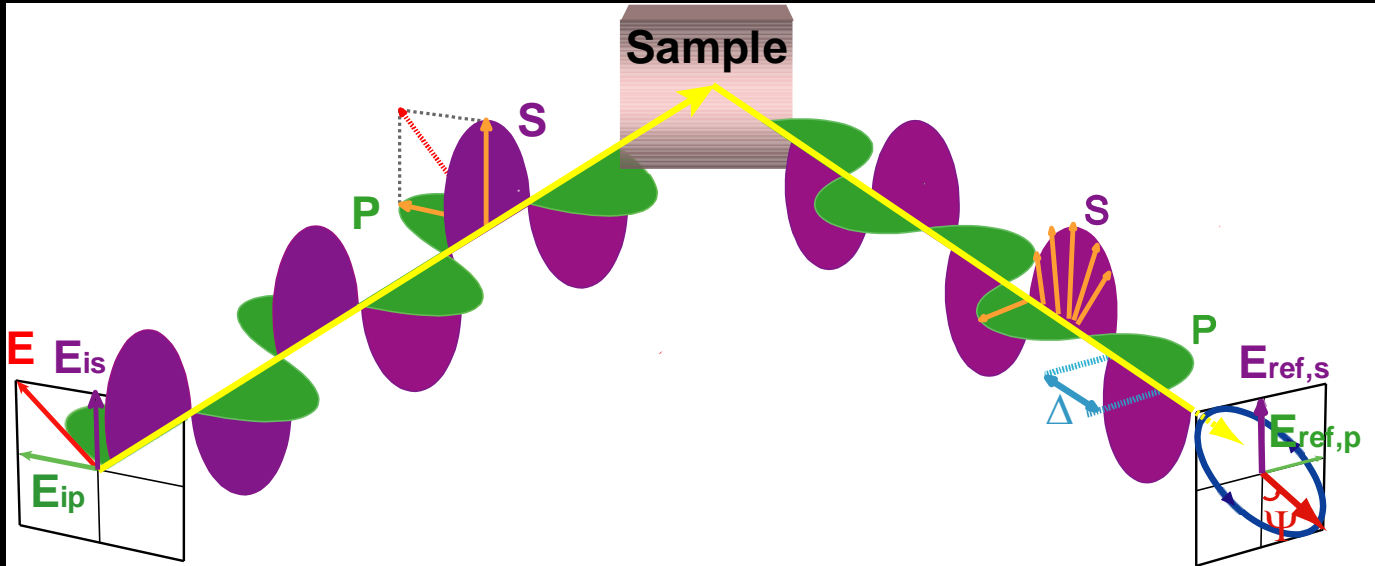
What can we measure with an Ellipsometer?

Changes in the polarization state of light when it is reflected from a surface of a sample

- ❖ The ratio of the amplitude reduction (Ψ)
- ❖ The phase shift induced by the reflection (Δ)

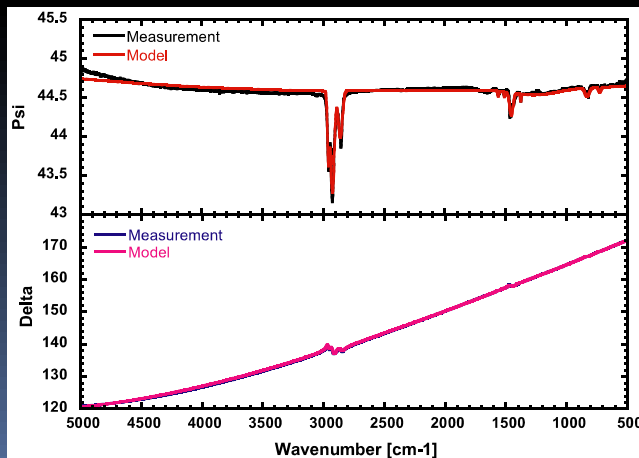


IR Ellipsometry -- Ψ & Δ --



$\Psi \rightarrow$ the ratio of the amplitude reduction (偏光の振幅比(屈折率))

$\Delta \rightarrow$ the phase shift induced by the reflection (偏光の位相差 (減衰係数))



➤ Ψ and Δ spectra of P3HT on an evaporated Au substrate prepared by spin coating (1000 rpm: concentration 2.5%)

➤ Measured with the incident angle of 60°

● Poly(3-hexylthiophene) (P3HT)

(thiophene ring with 6 carbons in the alkyl side chain)

● Type: p-type semiconductor

● Chemical formula: $(C_{10}H_{16}S)_x$

Ellipsometry -- Modeling --

These parameters are related to the **Fresnel reflection coefficient** R_p (p-polarized) and R_s (s-polarized). The fundamental equation for ellipsometry (the complex reflectance ratio ρ) can be described as

$$\rho = \frac{\tilde{R}_p}{\tilde{R}_s} = \tan(\Psi)e^{i\Delta}$$

- A pseudo-dielectric function can be derived by the measured values Ψ & Δ

$$\boldsymbol{\varepsilon} = \boldsymbol{\varepsilon}' + i\boldsymbol{\varepsilon}'' = \sin^2(\Phi_1) \left\{ \mathbf{1} + \tan^2(\Phi_1) \left(\frac{\mathbf{1} - \boldsymbol{\rho}}{\mathbf{1} + \boldsymbol{\rho}} \right)^2 \right\}$$

ε' : real part

ε'' : imaginary part

Φ_1 : the angle of incidence

Film Thickness

Film thickness can be determined by interferences between light reflected from different surfaces:

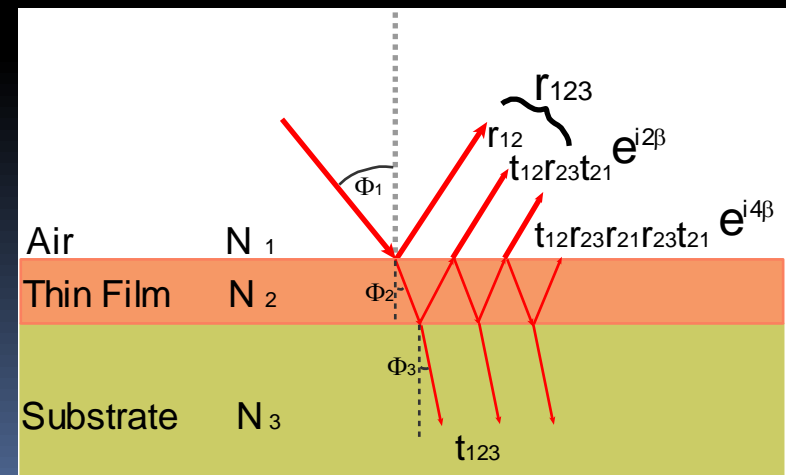
$$d = \beta \frac{\lambda}{2\pi \sqrt{N_2^2 - N_1^2 \sin^2(\Phi_1)}}$$

d : the film thickness

β : the phase factor

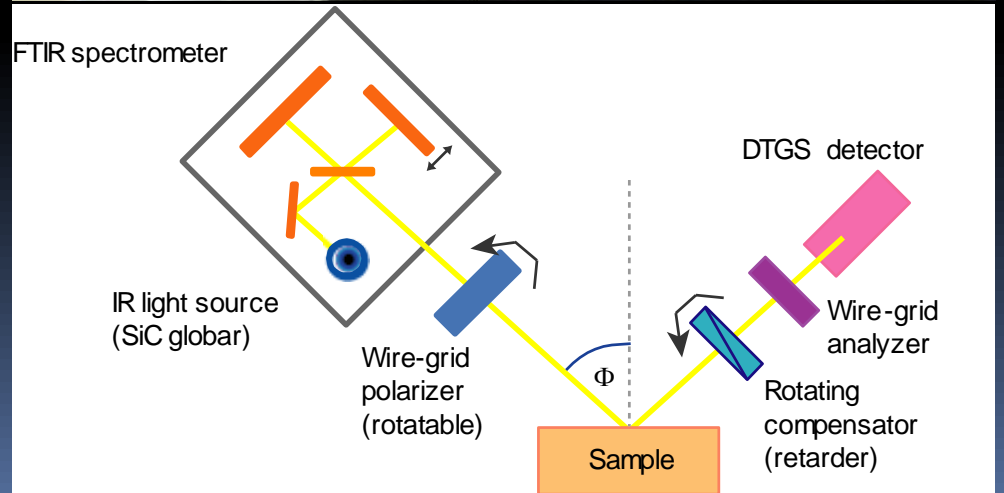
Φ_1 : the angle of incidence

λ : the wavelength of light



cont. IR Ellipsometry

- ☀ A rotating compensator IR Ellipsometer from J.A. Woollam Co., Inc (IR-VASE)
- ☀ In the spectral range of 333 to 5900 cm^{-1} (30 – 1.7 μm)
- ☀ The angle of incidence can be adjusted the range between 26° and 90°
- ☀ Ψ & Δ measurements



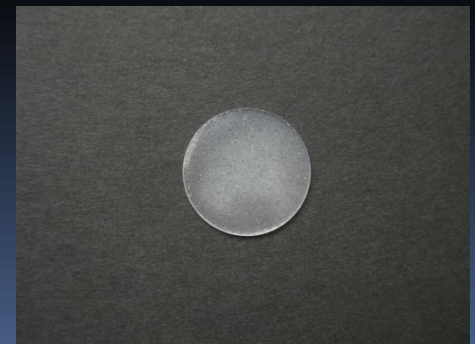
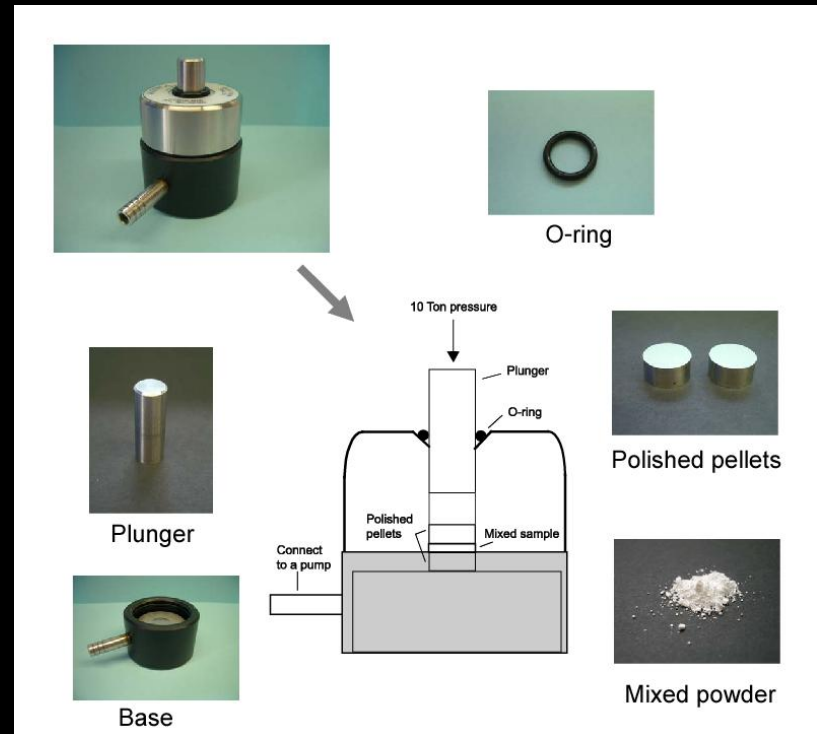
Sample Preparation

Depending on the experimental purpose and device, the sample preparation changes

- Pellet technique
- Aerosol spectroscopy
- On substrate
 - Evaporation
 - Spin coating

Pellet Technique

- The ratio of a sample and the KBr powder is 1 to 500 in weight (when a particle size is about $1\ \mu\text{m}$).
- Mixing them well in order to segregate the particle as minuscule as possible.
- For making a 0.55mm thick and 13 mm in diameter pellet, 0.2g of the mixed sample is placed in an evacuable pellet die.
- Using a hydraulic press, first evacuating all air slowly and applying 10 Ton load to the die for 20-30 min.
- As a medium
 - KBr ($0.23\text{-}25\ \mu\text{m}$)
 - NaCl ($0.18\text{-}20\ \mu\text{m}$)
 - CsI ($0.24\text{-}70\ \mu\text{m}$)



Pellet Technique -- Adv. & Disadv.--

Advantages:

- Low sample consumption
- Possible to keep the pellets in a desiccator
- Quantitative measurement is possible → Possible to calculate an absorption coefficient

$$\kappa \text{ (absorption coefficient)} = (\text{absorbance} \cdot \text{area of the pellet (cm}^2\text{)})/m$$

m: mass of the samples in the pellet (mg)


Disadvantages:

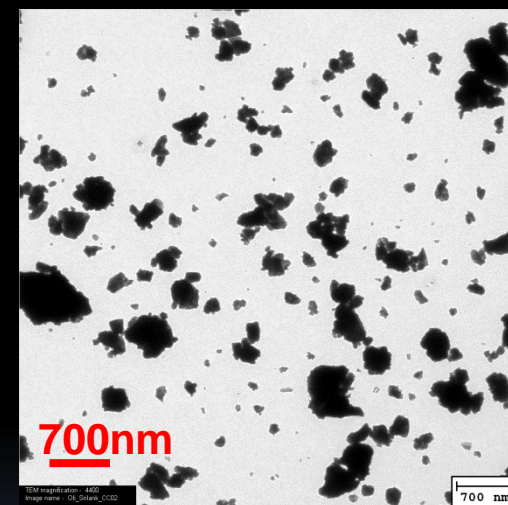
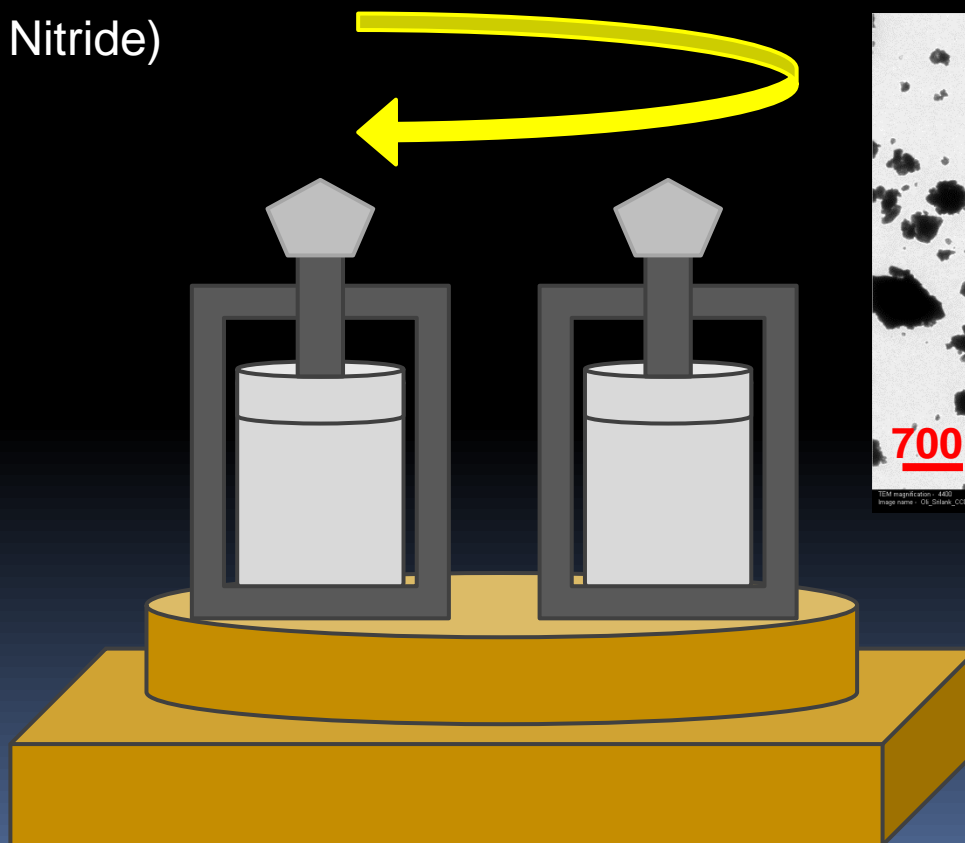
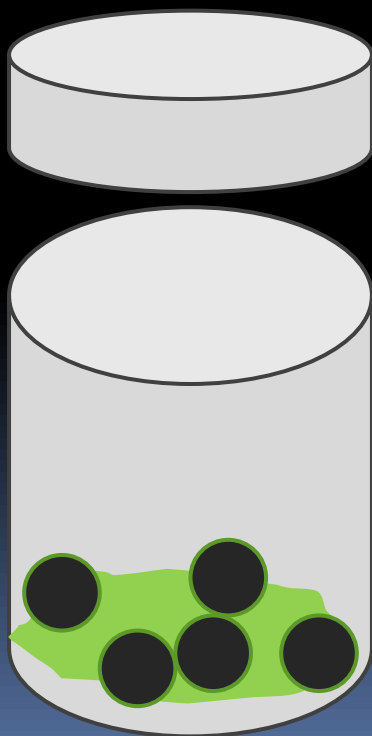
- Particles may transform during the grinding procedure
- A configuration of particles can hardly be visible whether particles are thoroughly isolated in the KBr matrix or not
- There is a risk to cause the powdered sample structure deformation in relation to high pressurization required for the technique

Samples for Aerosol Spectroscopy

Ball mill

→ A grinder --- possible to grind a material into micron-sized particles by making use of centrifugal force

 Si_3N_4 (Silicon Nitride)



TEM image
(Natural olivine)

On Substrate

Transmission:

- Si
- Al₂O₃
- Diamond ...

Reflection:

- ITO (indium tin oxide)
- Au
- Ge ...

Substrate thickness:

0.7 – 1 mm (Thicker → avoid the interference, but absorption gets stronger)

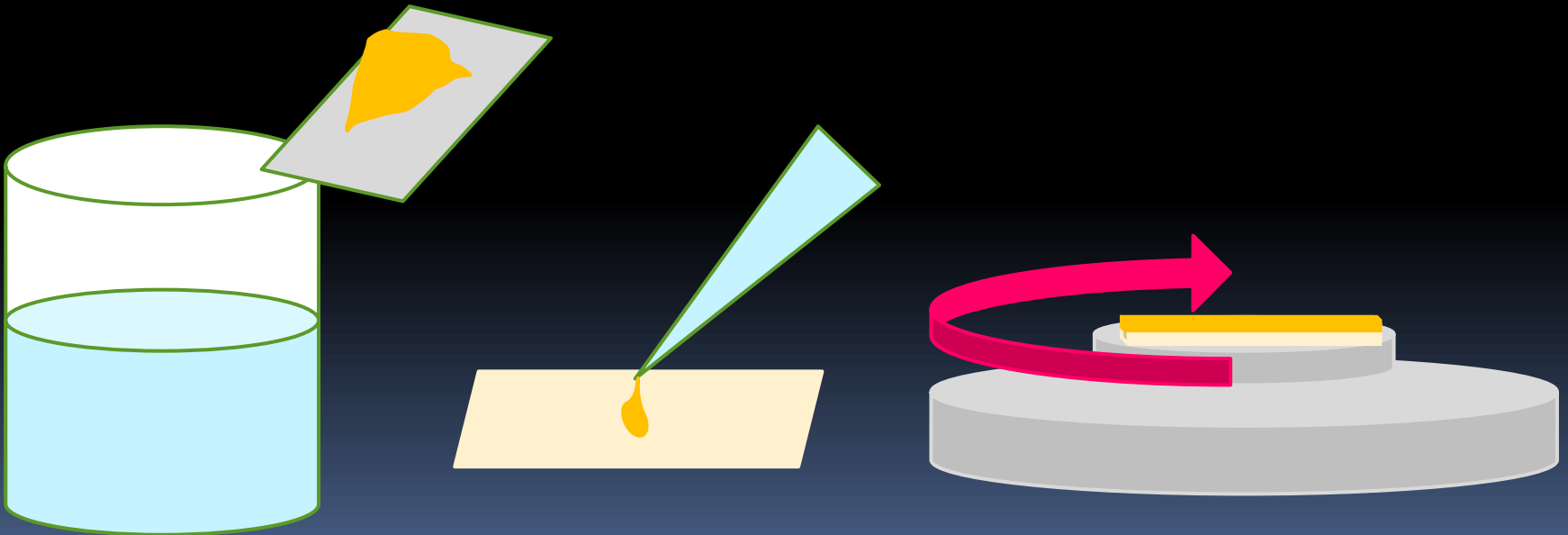
Vacuum evaporation:

- Clean the wafer with ethanol and acetone in order to remove the unnecessary water on the wafer
- Set the pressure 10^{-8} to 10^{-9} mbar and temperature 150°C for 2 to 3 hrs for cleaning the surface of the wafer
- After 30 min. cooling, transfer the wafer to the sample chamber
- Set the pressure 10^{-9} mbar and temperature around 130 to 150°C
- Switch on the filament for 30 min. Then, put the wafer in the sample chamber
- Bake out the wafer for 2 hrs
- The thickness of the layer is determined by the microbalances in the chamber
- After finishing the bake-out, it is possible to start the spectroscopic measurements for the prepared sample

cont. On Substrate

Spin coating:

- Powdered sample is dissolved in a solvent
- Put a small amount of solution on a substrate
- Transfer the substrate to a spin coating machine
- The substrate is rotated on the machine at high speed, so the solution is uniformly distributed due to centrifugal force
- A film thickness is controlled by a spin speed & concentration of a solute

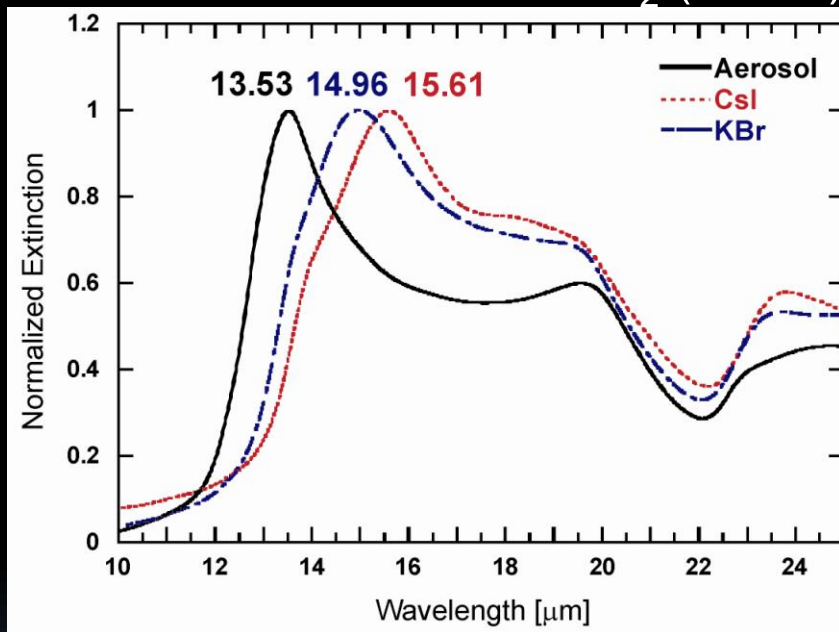


Measurements & Data Analysis

<Pellet vs. Aerosol Measurements>

--Medium effect--

TiO₂ (Rutile)



ϵ_m

N₂ → 1.0

KBr → 2.3

Csl → 3.0

The influence of its electromagnetic polarization.

(Tamanai et al. 2009)

KBr :

(Potassium Bromide)

Mixing ratio 1:500

(sample:KBr)

d=13mm ; mass=0.2g



Csl :

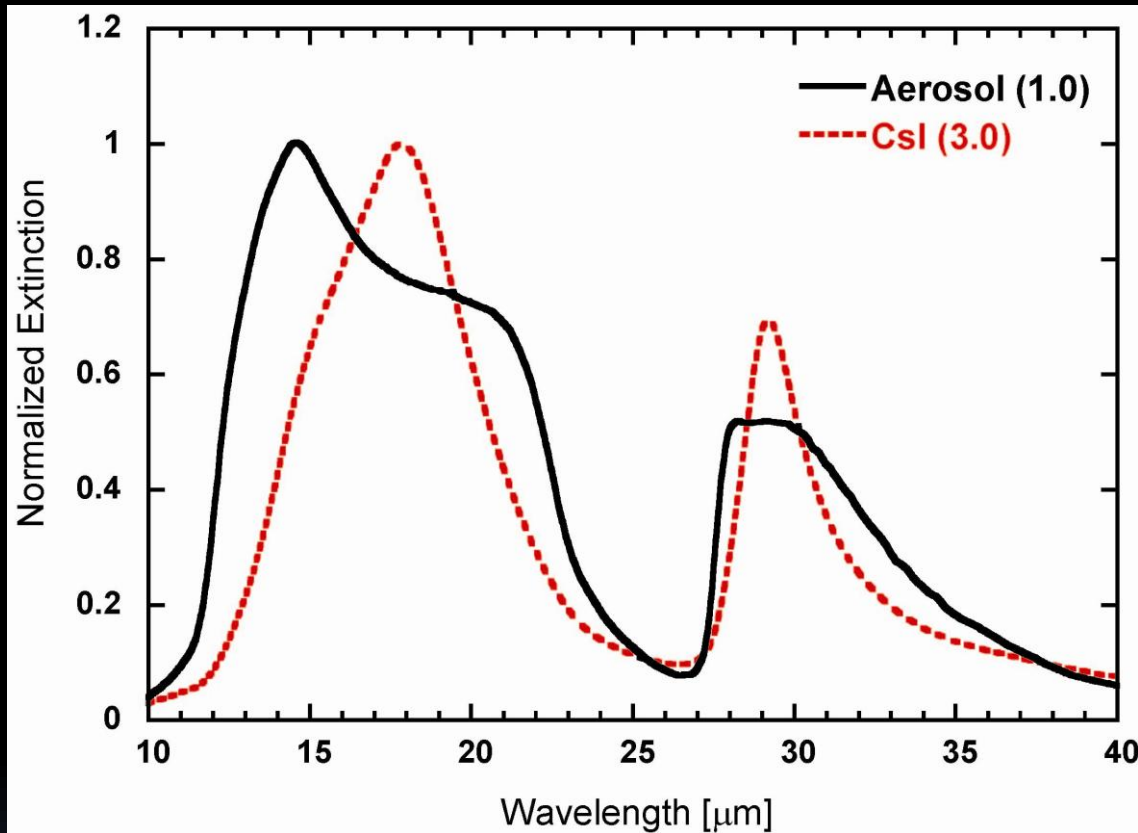
(Cesium Iodine)

Mixing ratio 1:500

(sample:Csl)

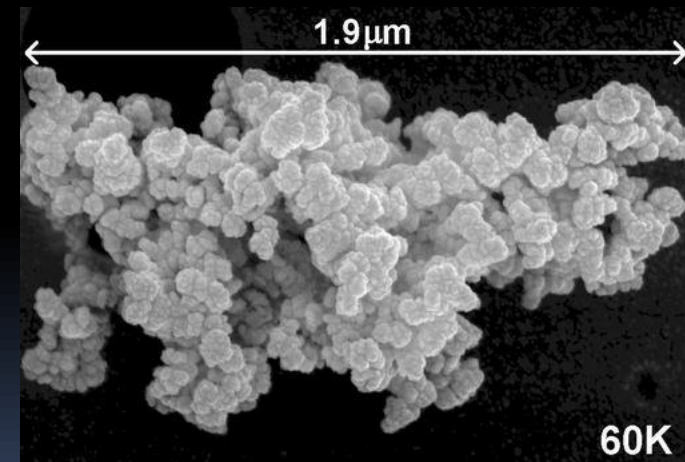
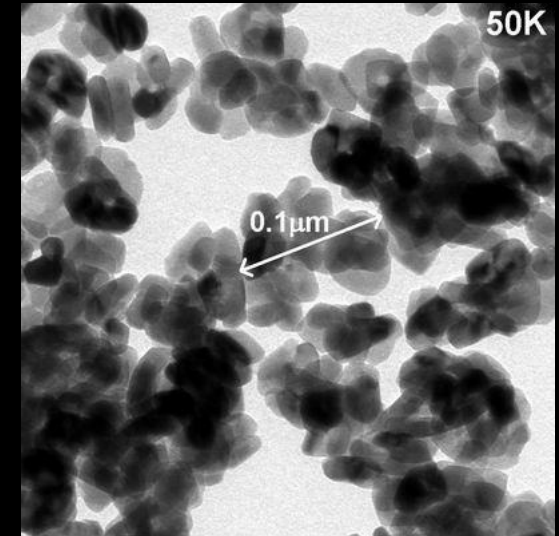
d=13mm ; mass=0.22g

-- Disparity in spectra --



Tamanai et al. (2009)

- Using different dispersion methods
- Particles may transform during the grinding procedure
- Sample structure deformation caused by the high pressurization required
- Particle orientation



-- Morphological effects --

TiO₂ (Rutile)

Irregular shape
w/ rounded
edges

Irregular shape
with not
rounded edges

Long and thin

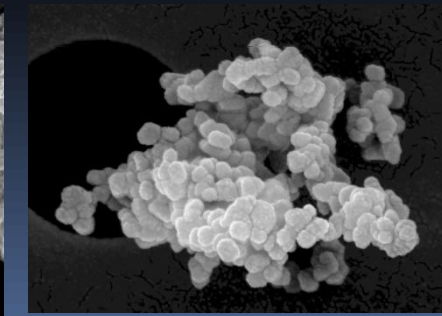
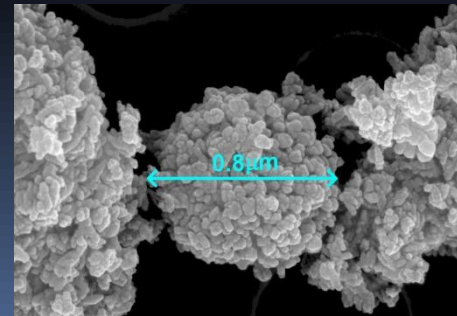
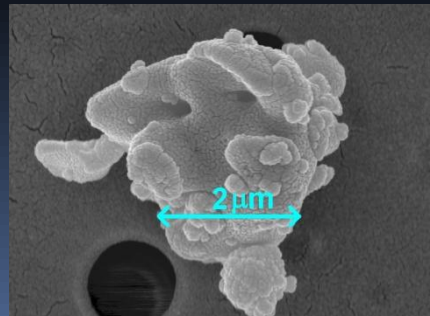
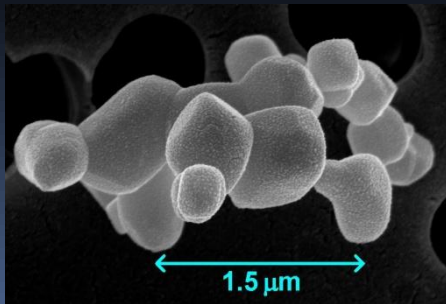
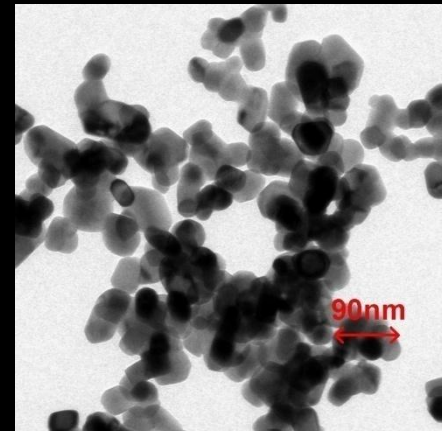
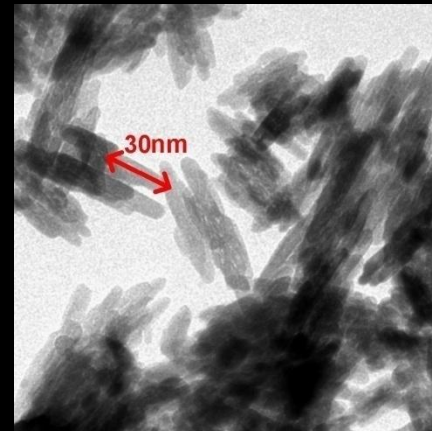
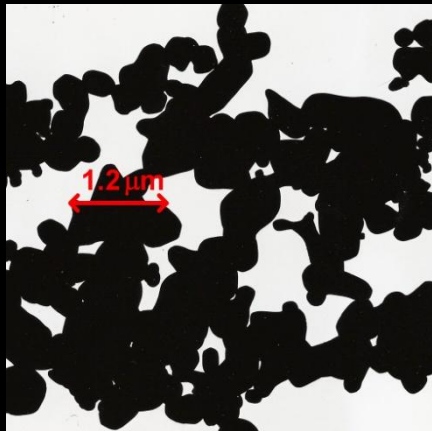
Rounded

R1

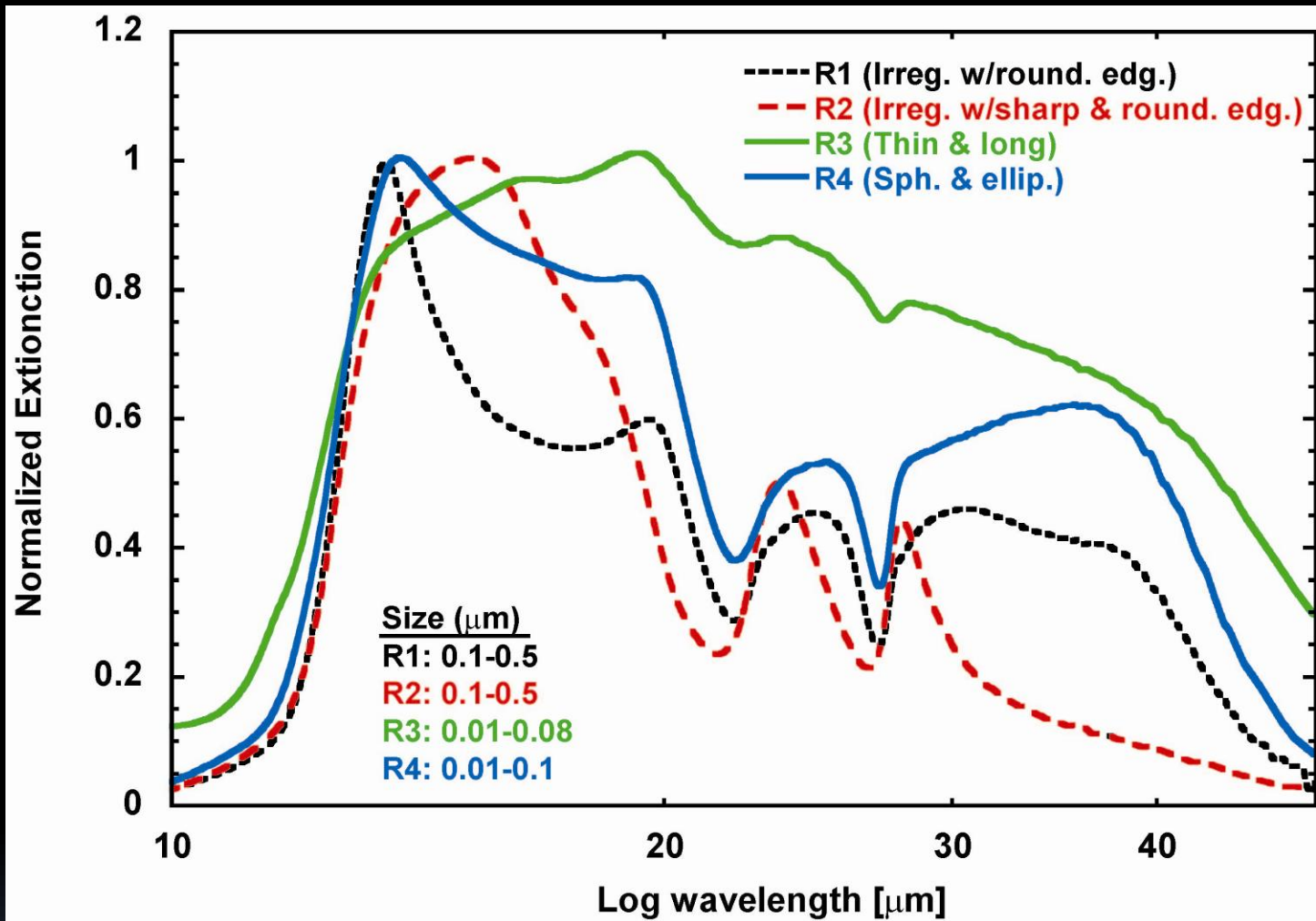
R2

R3

R4



(TEM & SEM images: at Pathology w/ Dr. Nietzsche)



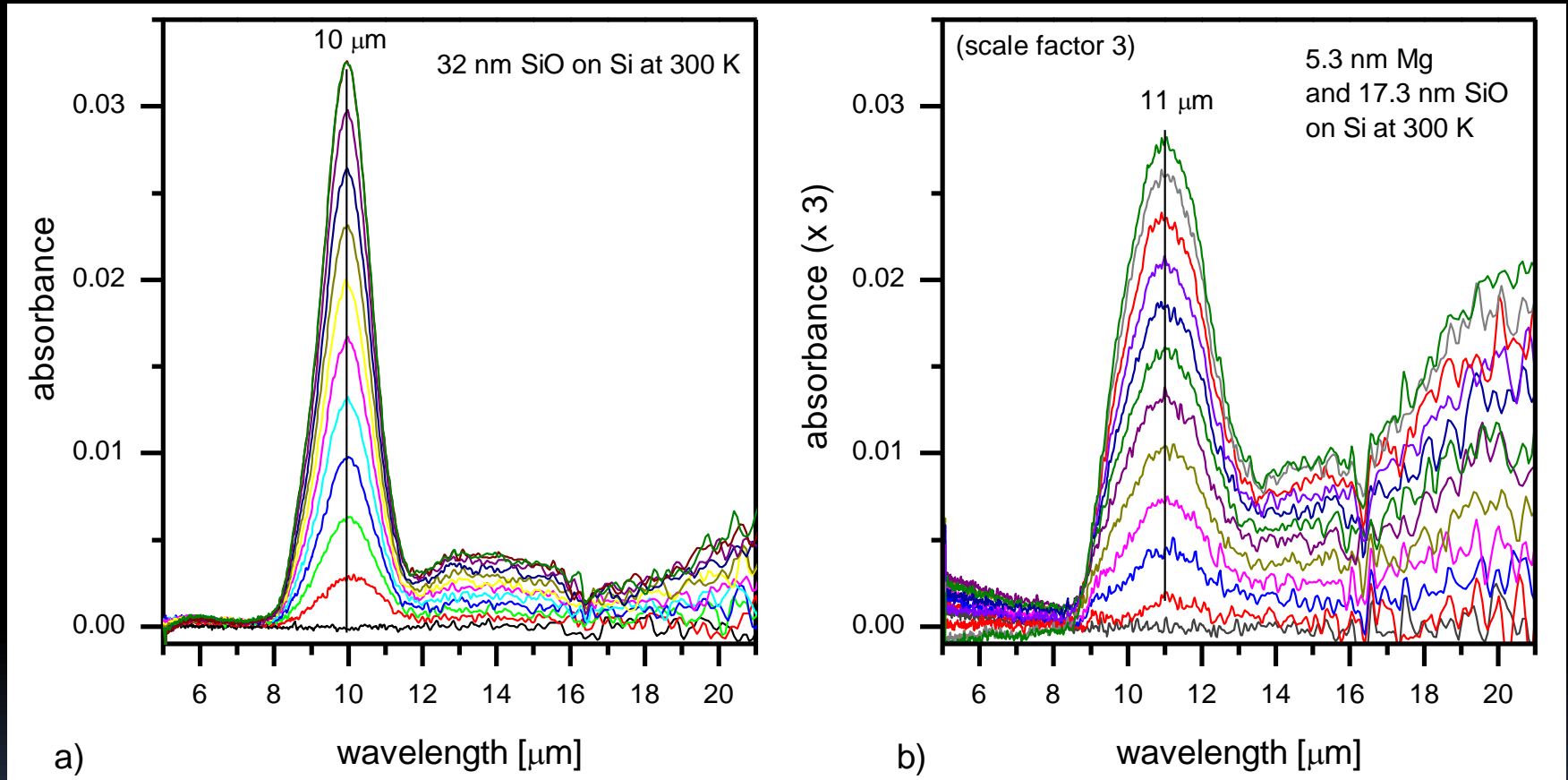
Peaks (microns)

R1	13.53	19.56	24.70	30.30	35.80
R2	15.33		23.51	27.95	
R3	16.47	19.25	23.39	28.27	
R4	13.85	19.27	25.15	35.87	

In situ IR Measurements

Co-evaporation – Mg and SiO

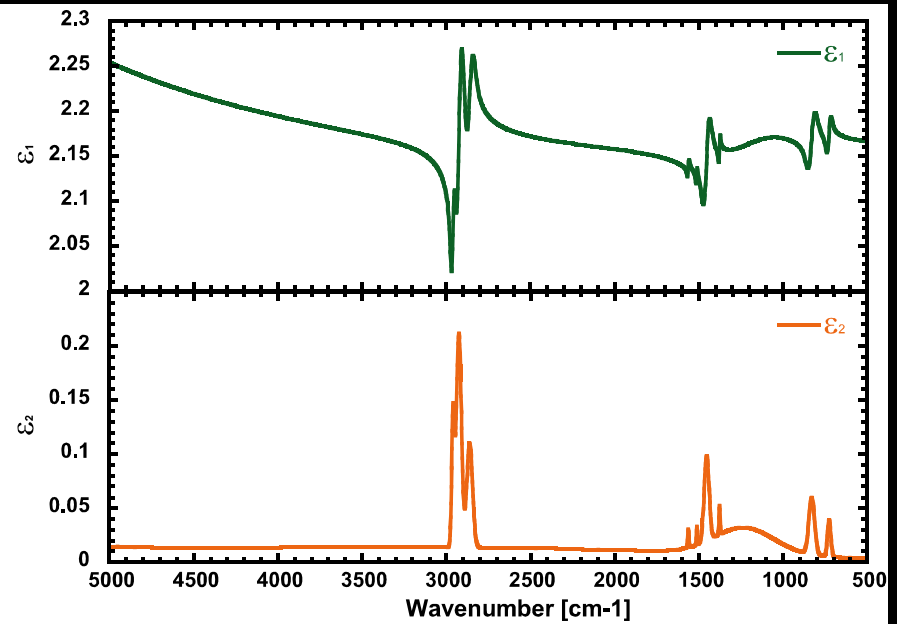
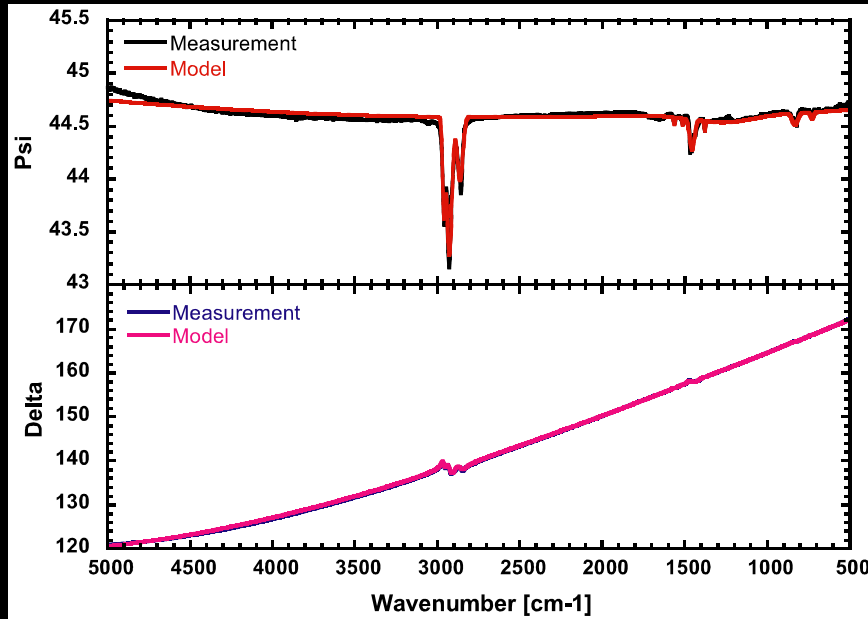
Measured by Wetzel (KIP)



- Redshift of the Si-O stretching vibration bands (from 10 μm to 11 μm) and broadening are confirmed. Metal itself seems to be “invisible”.

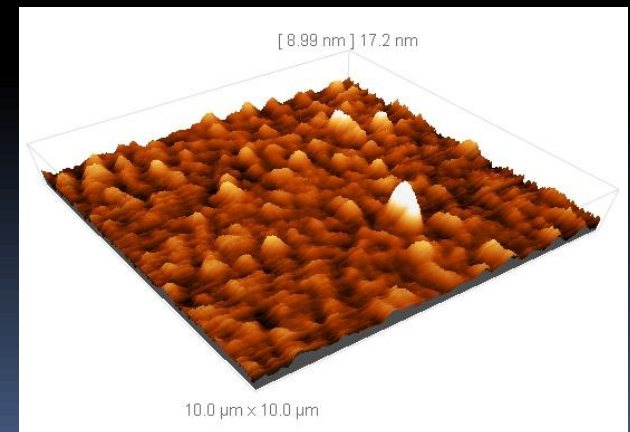
Ellipsometry Measurements

Spin coating (P3HT on Au substrate)



- The thickness of P3HT thin film deposited on Au substrate is 218.6 nm.
- The average surface roughness is approximately 15 nm obtained by AFM analysis (10x10 μm).
- All of the dielectric functions here is derived by means of Gaussian oscillators for model fittings.

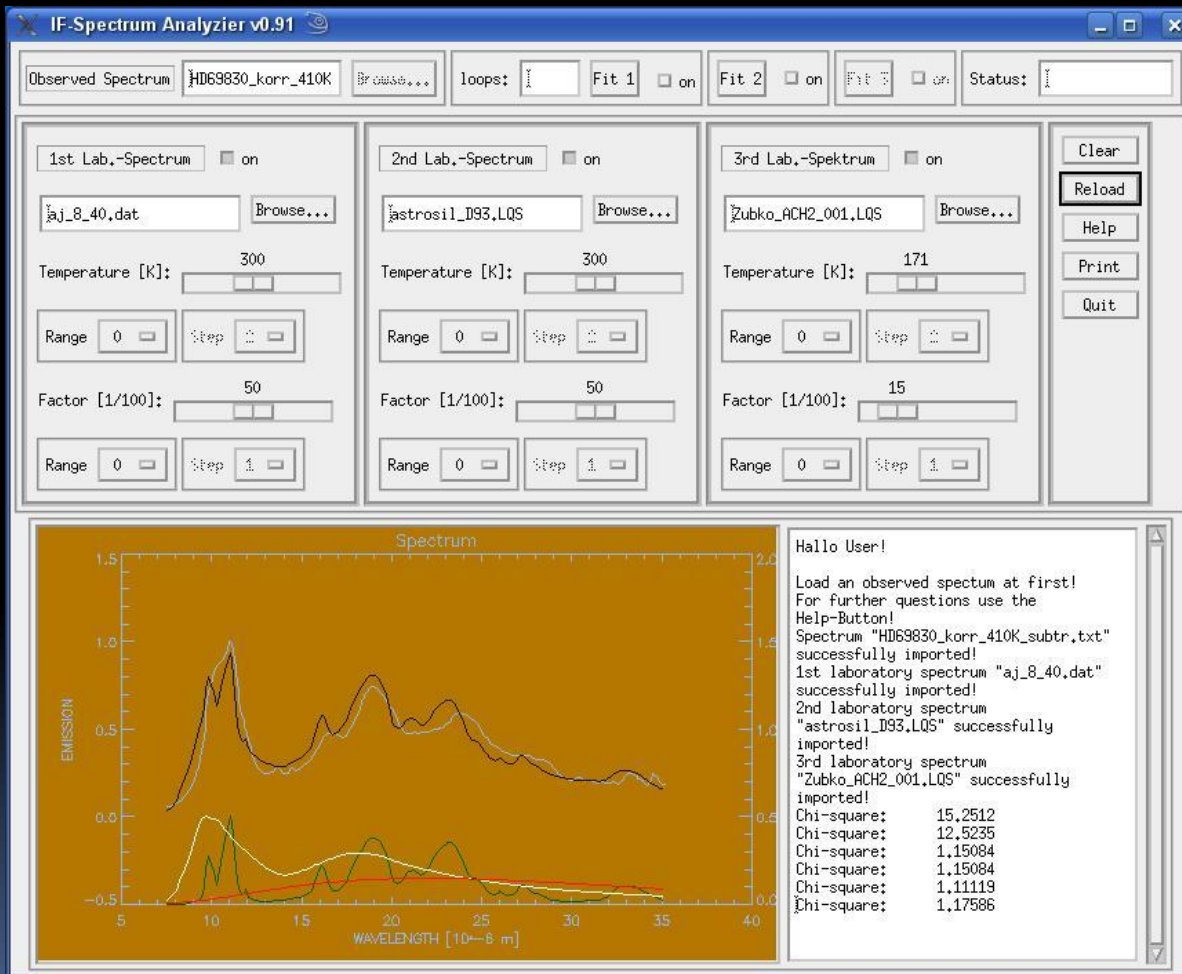
Tamanai et al. 2010 (conf.Tessaloniki, Greece)



Morphological feature of lateral dimensions

Astronomical Application

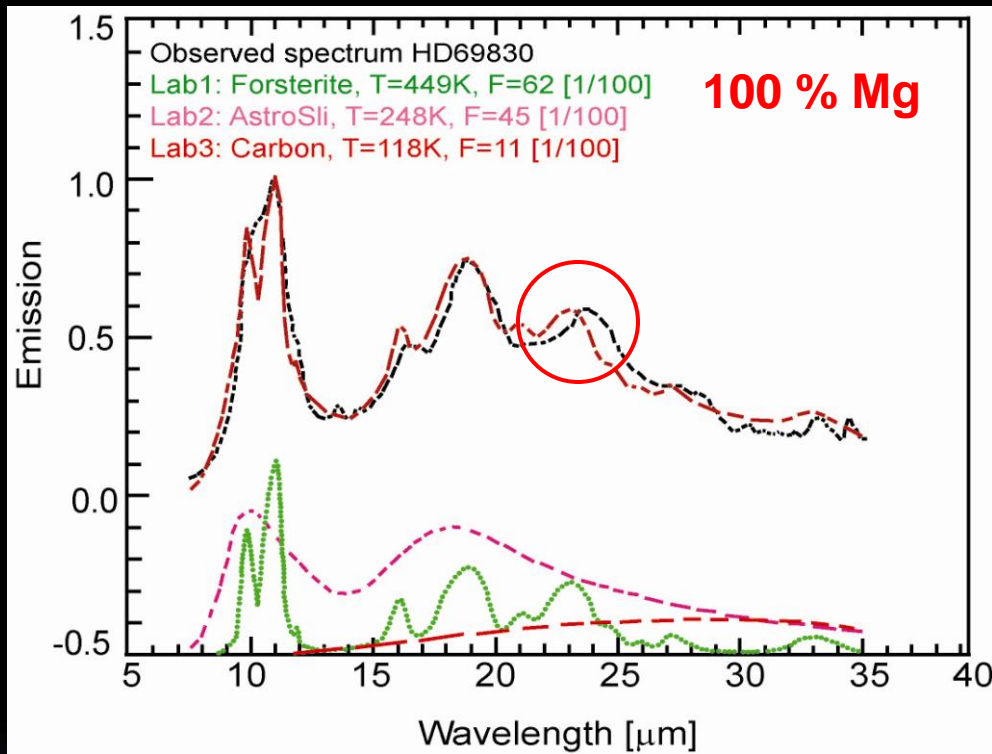
First steps: Some own fitting tests for HD 69830



➤ The spectrum → the product of the emission cross section of the dust with a Planck function given by the equilibrium T of the dust

– IDL programme IRSA by Lutz Bornschein

Some own fitting tests for HD 69830



- Beichman et al. (2005): Spitzer IRS
- $T(\text{dust}) \sim 400\text{K}$
- $M(\text{dust}) \sim 1000 \text{ zodi}$

K0 V
13 pc
2000 Myr ?
3 planets

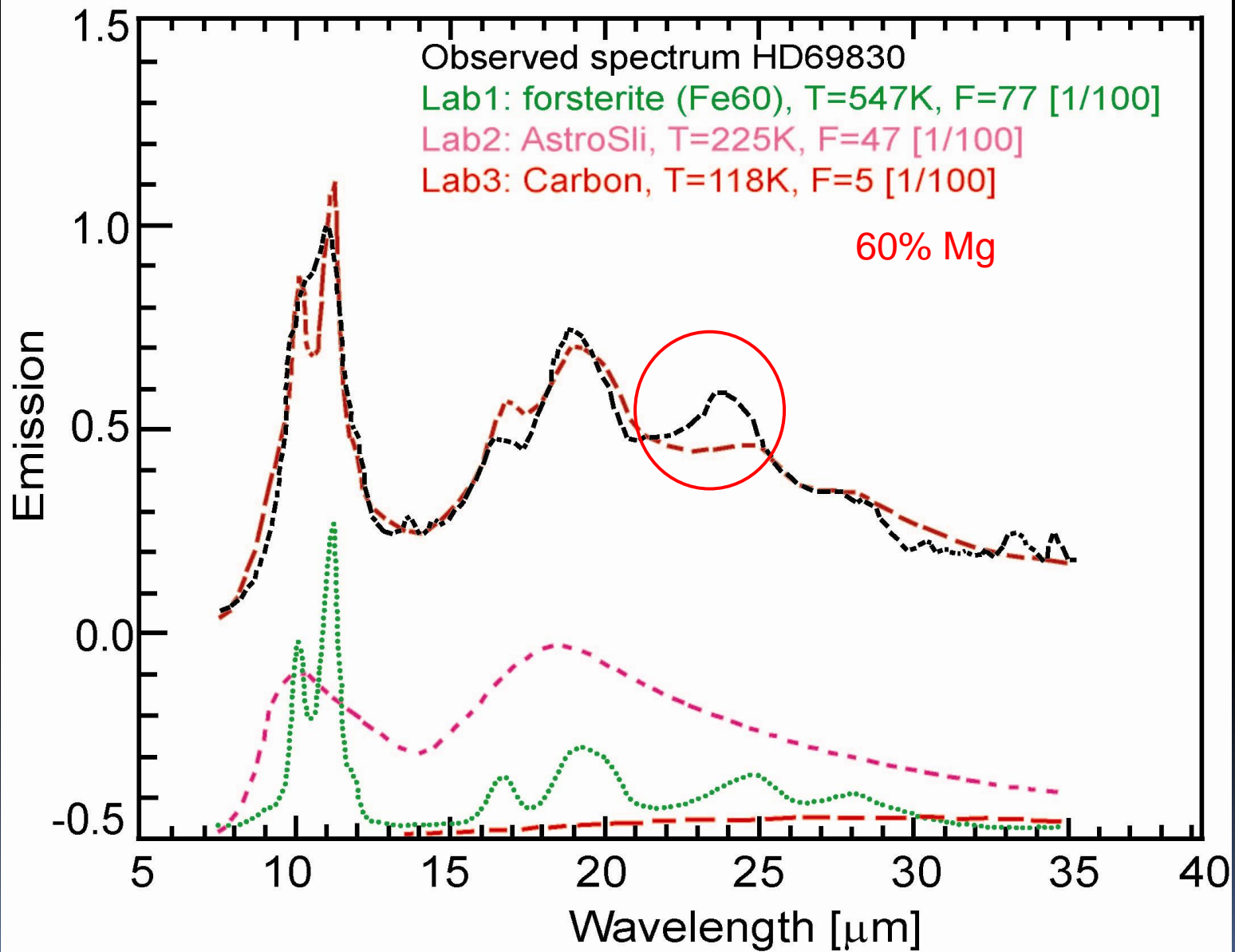
Intention: simplification, search for real „hard facts“ (constraints on dust mixture)

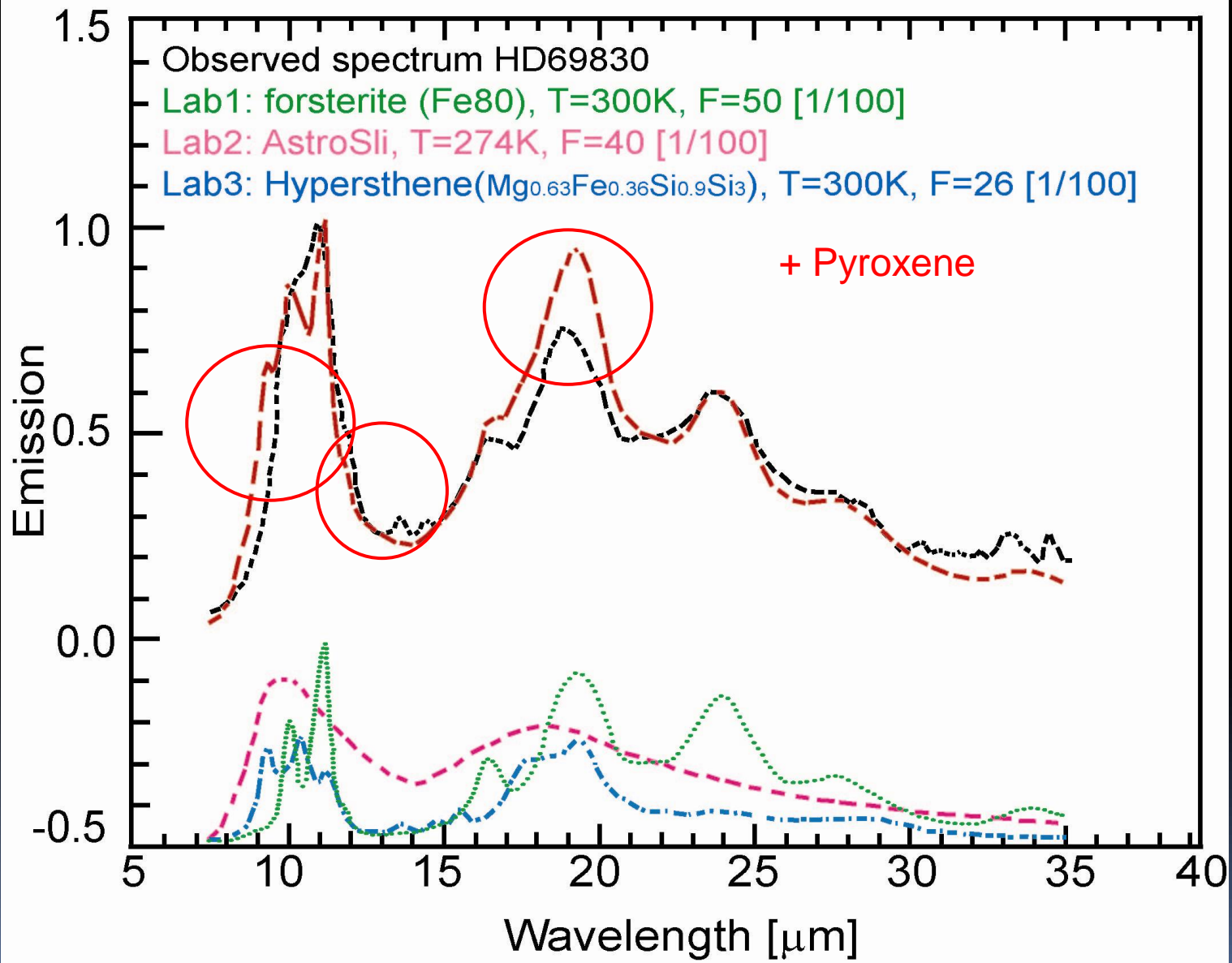
Used dust spectra:

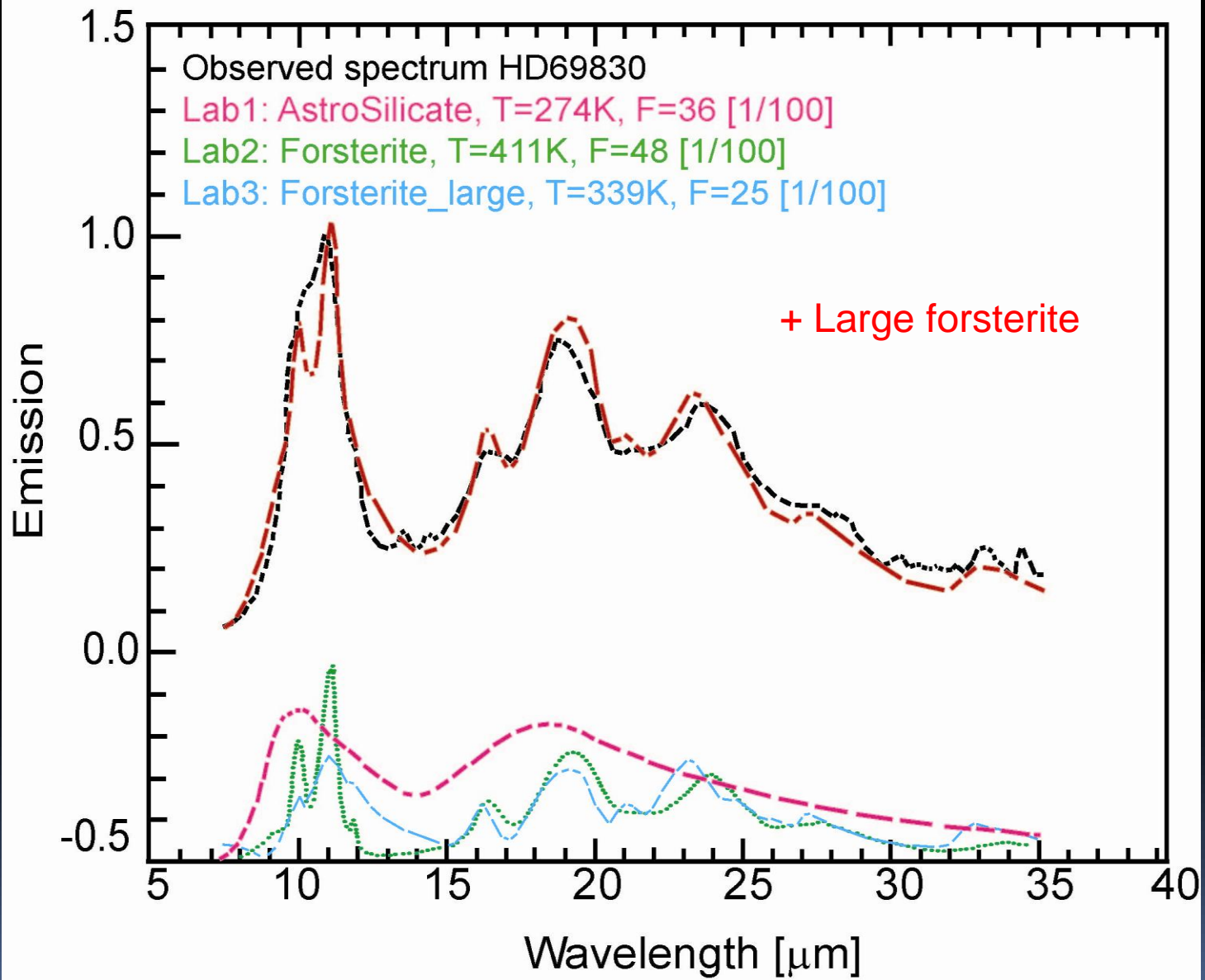
- olivines and pyroxenes from aerosol measurements
- Koike et al. (2003) olivines
- amorphous silicates
- „carbon“ for grey emission (may also mimic colder dust component)

Three-component model:

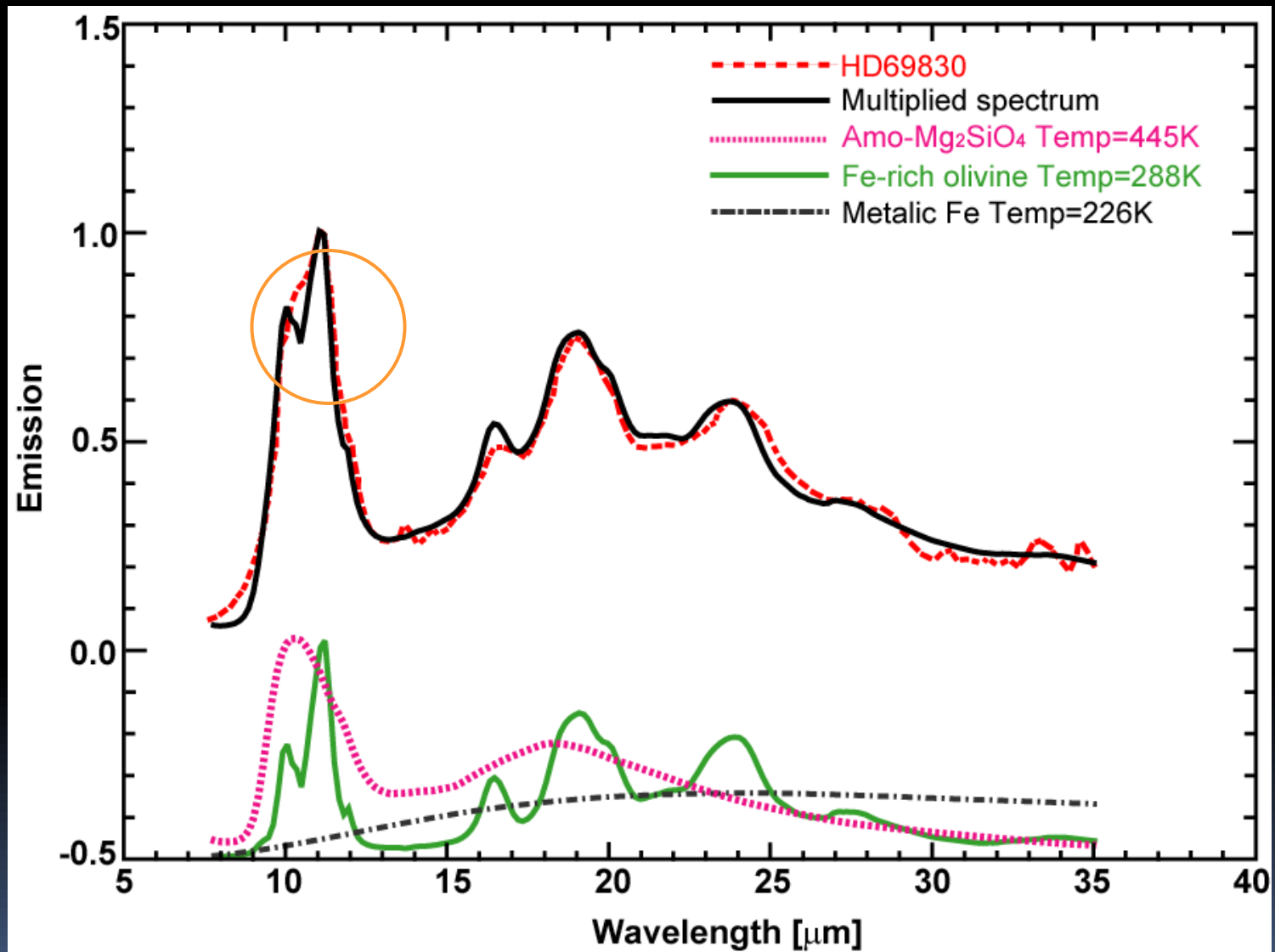
- Fo60+ olivine (synthetic, Koike)
- Astronom. Silicate (amorphous, D&L93)
- „carbon“ cold dust







Best fit for HD69830



Database

Peak Search: to search for your desired spectra via inputting a peak position.

Basic information: Chemical formula, density, particle size, product info., etc...

Plot: Aerosol and Csl pellet measurements. Numerical data sets are downloadable.

Images: obtained by scanning and transmission electron microscopes (SEM & TEM)

Database of aerosol spectra for cosmic dust
Laboratory Astrophysics Group of the AUL Jena

Navigation: Home > FSC > AIC > Database > CsI > Results > IO₂

Database
CsI
Spectral
Peaksearch
Sitemap
Contact

IO₂

Basic Information

Classification: Crystalline	Size: $d < 1 \mu\text{m}$
Chemical formula: IO ₂	Shape: Irregular poly-crystalline edges
Product Info: Aldrich	Density: 4.25 g/cm ³
Preparation: Origin	Melting Point: 242.1 °C @ 27
Reference:	

Plot

Wavelength [μm]	Q _{ext} (Q/V) [μm ⁻¹ /peak]
13.53	~0.9
15.81	~1.0
19.56	~0.5
23.77	~0.5
24.70	~0.5
28.13	~0.4
30.30	~0.4

Images

SEM: Magnifikation 90x
SEM: Magnifikation 25x
TEM: Magnifikation 1.5 μm
TEM: Magnifikation 570x

© 2011 Astro-Physik, Universität Jena

Possibilities for Future

Current project

Investigation for electro organics

Infrared spectroscopic measurements for organic thin films in ultra high vacuum (UHV) condition

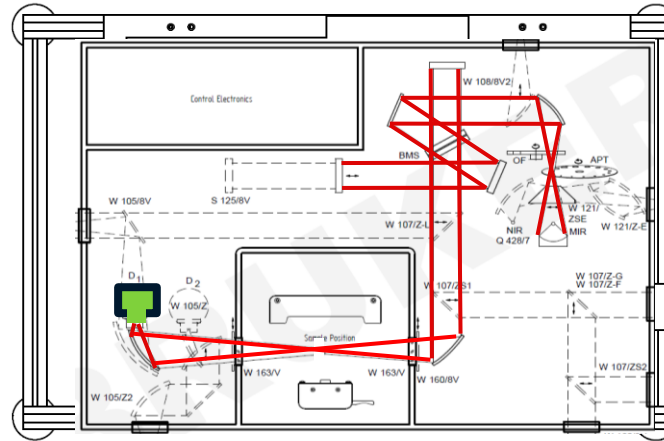
Verify:

- ▶ Effects through the surface metallization and interface intermixing
- ▶ Degradation of IR and VIS spectra by light, adsorbates, heating
 - Detect weak points & physical origin
 - Role of morphology and interfaces

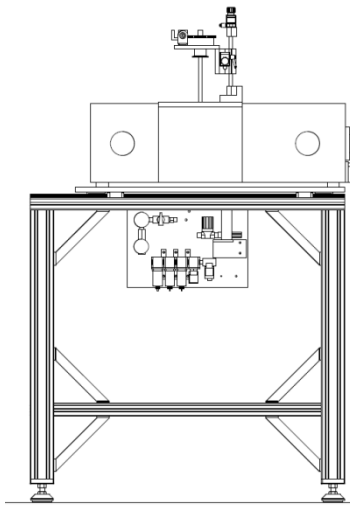
IR-Setup at Clustertool

BRUKER Vertex 80v FTIR spectrometer

Top view



Side view

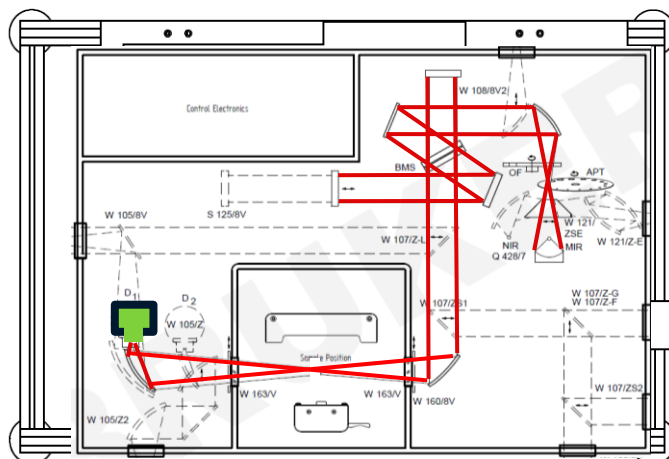


Diagrams by Glaser (KIP)

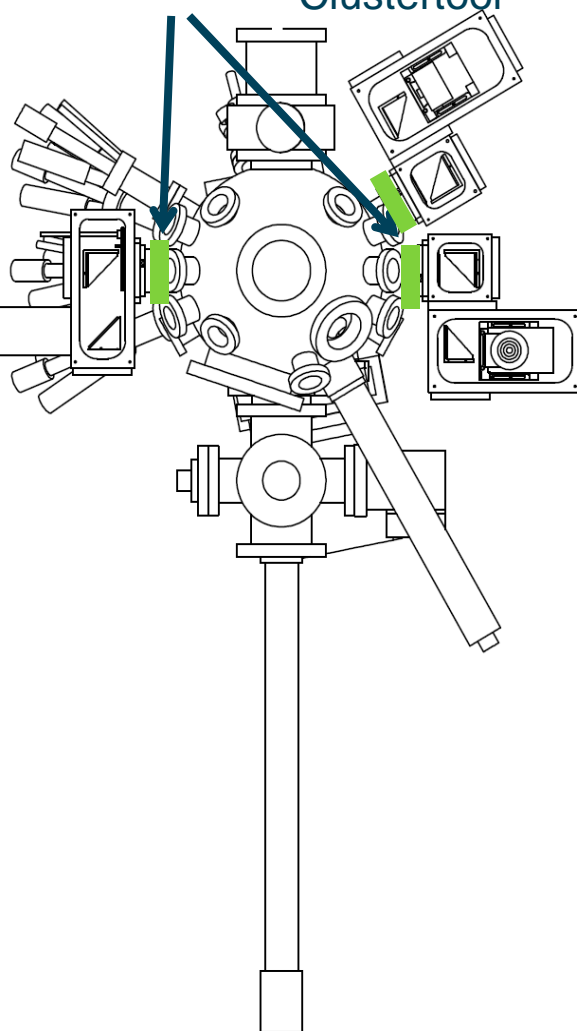
IR-Setup at Clustertool

BRUKER Vertex 80v FTIR spectrometer

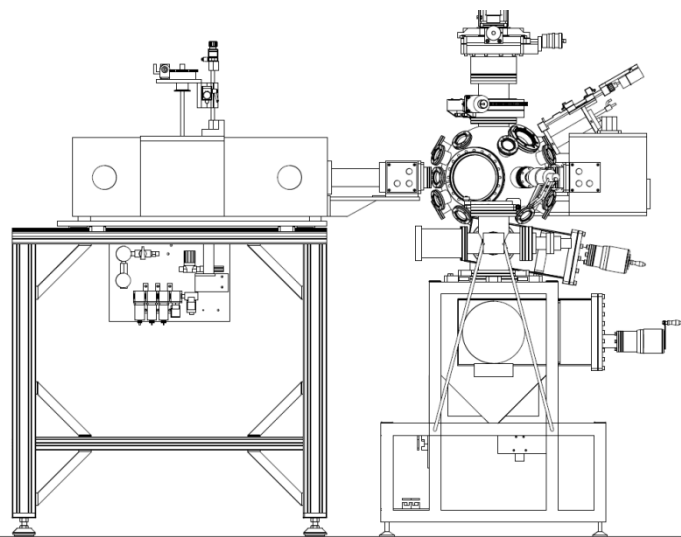
Top view



KBr window
Connection to
Clustertool



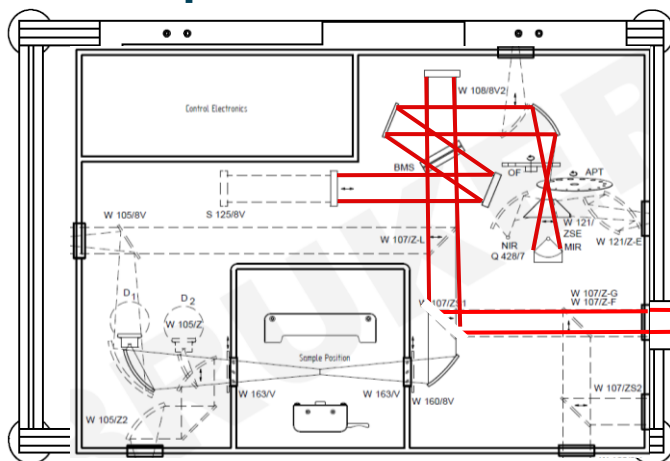
Side view



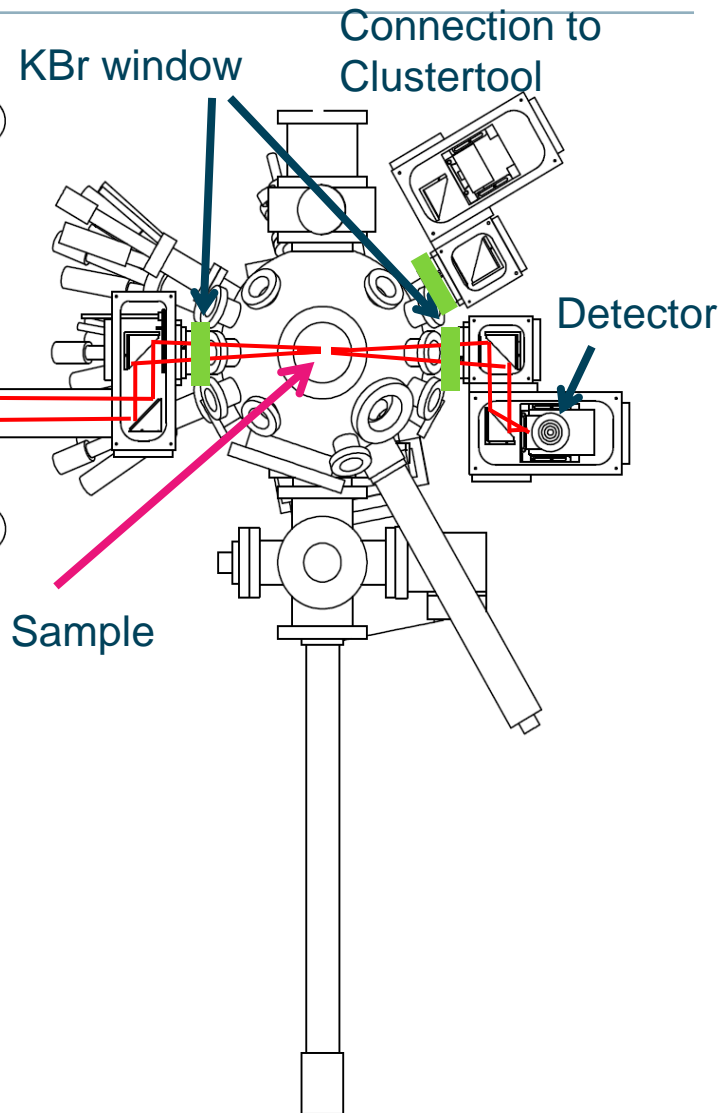
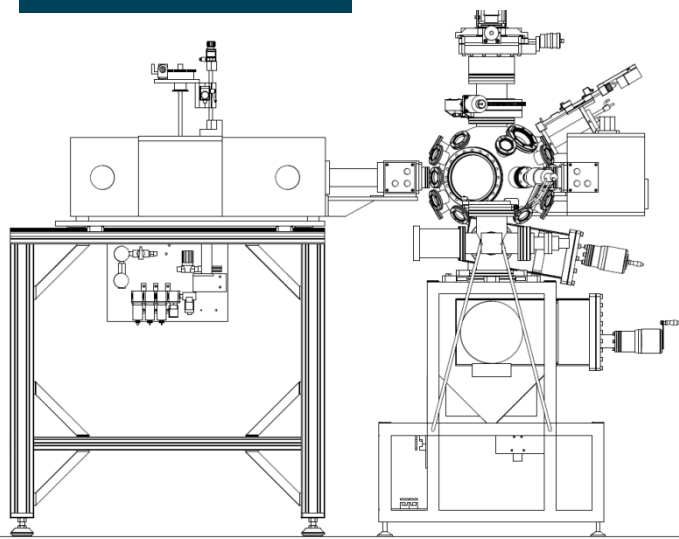
IR-Setup at Clustertool

BRUKER Vertex 80v FTIR spectrometer

Top view

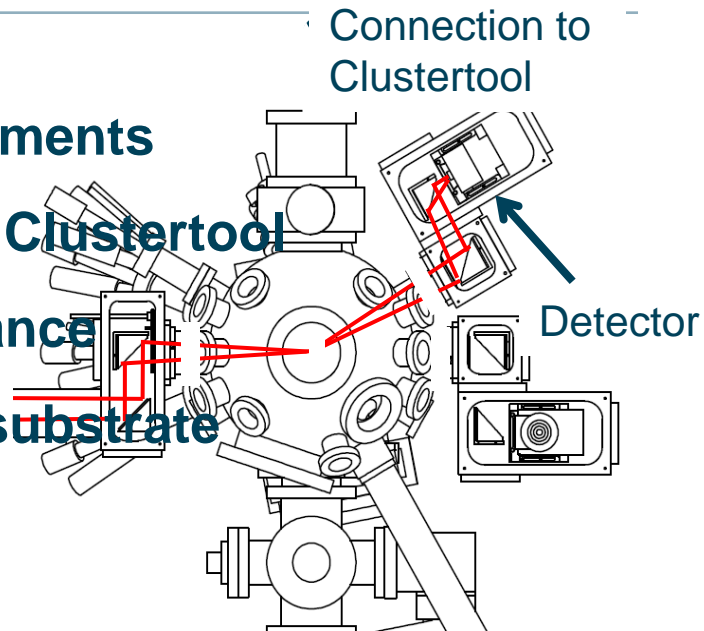


Side view



IR-setup at Clustertool

- Transmission and Reflection (75°) measurements
- Transfer from/to the other chambers of the Clustertool
- 2-crucible evaporator and quartz microbalance
- Resistive heating and cooling (LN₂) of the substrate
- Mass spectrometer
- Separate transfer system



- Spectral range extension for the FIR and NIR-range



Spectral range in spectrometer: ~ 20 – 10 000 cm⁻¹ (500 - 1 μm)

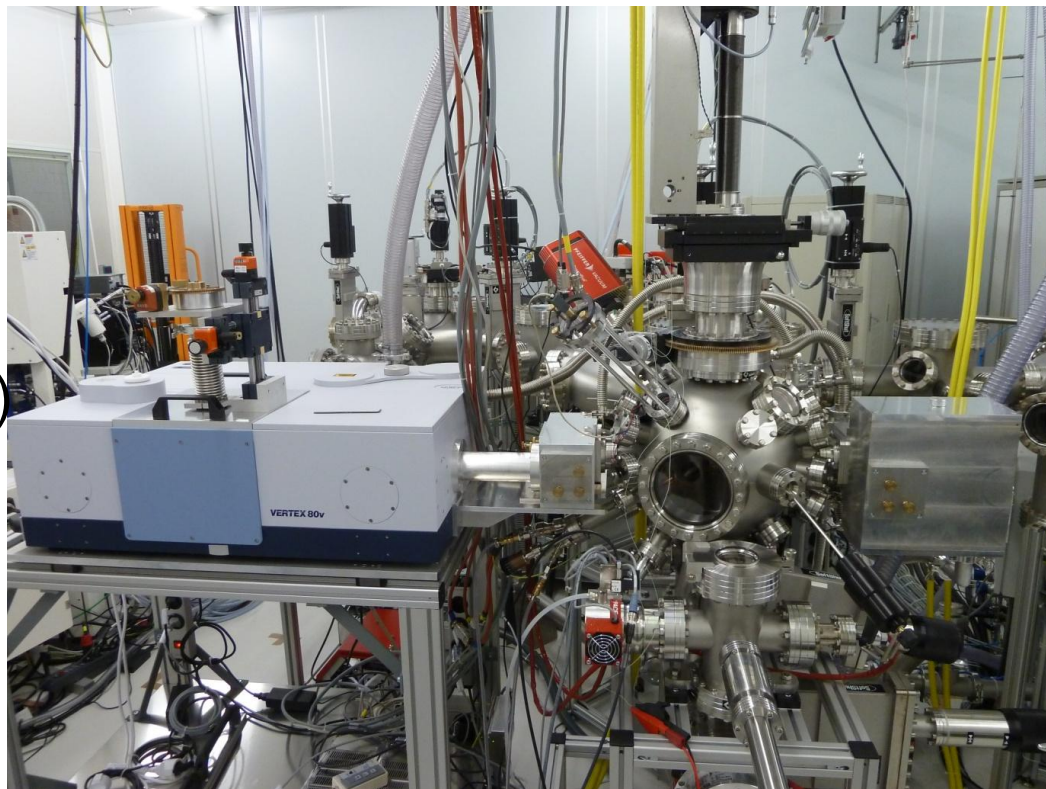
spectral range in UHV: ~ 330 – 10 000 cm⁻¹ (30 - 1 μm)

- UV-VIS-spectrometer

IR-setup at Clustertool

Devices:

- FTIR Spectrometer
- XPS (X-ray photoelectron spectroscopy)
- UPS (ultraviolet photoelectron spectroscopy)
- IPES (Inverse photoemission spectroscopy)
- AFM (Atomic force microscope)
- SEM w/ FIB (Scanning electron microscope with focused iron beam)



iL at Heidelberg, Germany

Summary

FTIR spectroscopy

- Aerosol spectroscopy
- In-situ FTIR spectroscopy
- IR ellipsometry

Sample preparation

- Pellet technique
- Aerosol technique
- On substrate

Current to future

- The growth of organic layers under physical vapor deposition
- Optical properties
- Interface effects

e.g.

Deriving optical constants of deposited film by means of IR ellipsometry

→ Derived optical constants can be used for theoretical calculations

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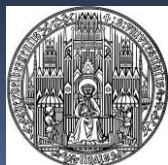
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